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Journal of the European Ceramic Society 30 (2010) 1317-1325

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Effect of TiO₂ doping on the characteristics of macroporous Al₂O₃/TiO₂ membrane supports

Hong Qi^{a,*}, Yiqun Fan^a, Weihong Xing^a, Louis Winnubst^b

^a State Key Laboratory of Materials-Oriented Chemical Engineering, Membrane Science and Technology Research Center, Nanjing University of Technology, No. 5 New Model Road, Nanjing 210009, China

^b Inorganic Membranes, Membrane Technology Group, MESA⁺ Institute for Nanotechnology, University of Twente, P.O. Box 217, 7500 AE Enschede,

The Netherlands

Received 2 August 2009; received in revised form 17 November 2009; accepted 17 December 2009 Available online 12 January 2010

Abstract

A cost-effective tubular macroporous ceramic support consisting of alumina and titania was prepared by extrusion and subsequent heat treatment. An Al_2O_3/TiO_2 composite support with high porosity (41.4%), an average pore size of 6.8 µm and sufficient mechanical strength (32.7 MPa) was obtained after sintering at 1400 °C. The formation mechanism of this support as investigated with X-ray micromapping, SEM and XRD indicated that the appearance of Al_2TiO_5 plays a key role in the fabrication of high performance composite membrane supports at relatively low temperature. The amount of Al_2TiO_5 present in the composite has a strong impact on the properties of supports, especially with regard to the mechanical strength. A composite of 85 wt.% $Al_2O_3/15$ wt.% TiO₂ sintered at 1400 °C for 2 h exhibited both high permeability (pure water flux of 45 m³ m⁻² h⁻¹ bar⁻¹), together with an excellent corrosive resistance towards hot NaOH and HNO₃ solutions. © 2009 Elsevier Ltd. All rights reserved.

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Keywords: Ceramic membranes; Macroporous support; Al₂O₃; TiO₂; Al₂TiO₅

1. Introduction

Since their introduction to commercial applications in the early 1980s, owing to their thermal, mechanical and chemical stability, ceramic membranes are making rapid progress in many areas such as food and beverage processing, biotechnology applications and water treatment.^{1,2} However, the disadvantages such as high cost and low surface area to module volume ratio³ greatly restrict its broad application. Ceramic membranes generally consist of several thin separation layers with thicknesses between a few tens of nanometers up to a few microns superimposed on a macroporous support. The support provides mechanical strength to the top membrane layer and must simultaneously have a high permeation and a high corrosion resistance to the filtrate flow.⁴ The high costs of ceramic membranes are partly attributed to the macroporous ceramic support, which is prepared via extrusion with subsequent drying and sintering processes. To impart the membrane support with sufficient

0955-2219/\$ - see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.jeurceramsoc.2009.12.011 mechanical strength as well as high permeability, most of the commercial macroporous ceramic supports (e.g. Membralox[®], Cefilt[®]) were made from alumina powders with particle size in the range of 20–40 μ m. Because of the extremely low sintering activity of these coarse-grained alumina, no sufficient strength could be obtained unless the sintering temperature reaches 1700 °C, or even higher.⁵

In order to fabricate these types of ceramic membrane supports at low sintering temperatures, several methods were proposed in the last 2 decades. A general method was to fabricate with starting powder of cordierite,⁶ mullite⁷ or a composite of alumina/clay,^{8,9} whose sintering temperatures were all much lower than that of pure alumina.^{10,11} To modify the powders with improved sintering activity so as to reduce the sintering temperature of the membrane support was also put forward as an alternative approach.^{12,13} Nevertheless, in order to design membrane supports, one must keep in mind the integrative properties, i.e. sufficient permeability as well as mechanical and chemical stability.¹⁴ Most of the studies with respect to membrane support are focused on either permeability or mechanical property, while one of the most important parameters, the corrosion resistance of the support was seldom studied. Therefore, inventive solutions in

^{*} Corresponding author. Tel.: +86 25 83172279; fax: +86 25 83172292. *E-mail address:* hqinjut@yahoo.com.cn (H. Qi).

Table 1	
Chemical analysis of alumina and titania powders.	

Oxide	Alumina (wt.)	Titania (wt.)
SiO ₂	350 ppm	0.3%
Fe ₂ O ₃	300 ppm	200 ppm
MnO	<100 ppm	<100 ppm
MgO	<100 ppm	200 ppm
CaO	100 ppm	200 ppm
Na ₂ O	0.19%	0.25%
K ₂ O	250 ppm	200 ppm

designing commercial low-cost macroporous ceramic supports with high performance (integrated properties of permeability, mechanical and chemical stability) are still in progress.^{7,9}

 α -Al₂O₃ and TiO₂ are normally used as membrane materials because of their excellent chemical stability towards extreme pH conditions. More important, it was reported¹⁵ that alumina and titania could form aluminum titanate through solid-state reactions above 1280 °C. The appearance of aluminum titanate in the Al₂O₃/TiO₂ system increases the concentration of vacancies and accordingly the velocity of mass transfer.^{16,17} The required properties for membrane supports such as sufficient mechanical strength as well as high permeability could be simultaneously obtained at a remarkably low temperature. Although the fabrication of porous alumina/titania membrane supports has been reported,^{18–20} few studies are available with respect to the formation mechanism of the Al₂TiO₅ phase and the effect of sintering temperature on support properties.

In this paper we report the microstructural characteristics and mechanical properties of tubular macroporous ceramic membrane supports consisting of alumina and titania made by extrusion while sintered at a comparative low temperature.

2. Experimental

2.1. Characterization of the starting powders

 α -Al₂O₃ (purity > 99.5%, Zhengzhou, China) and TiO₂ (purity > 99%, Nanjing, China) powders were used as received. Table 1 shows the chemical analysis (determined by sequential X-ray fluorescence spectrometer, XRF, VF-320, Shimadzu, Japan) of these starting powders. Fig. 1 depicts the alumina and titania particle size distributions as measured by dynamic light scattering (DLS) using Mastersizer 2000 apparatus (Malvern Instrument Co., Ltd., UK). The average size of α -Al₂O₃ and TiO₂ was ~30 µm and ~0.5 µm, respectively.

2.2. Preparation of the tubular macroporous supports

The above-mentioned Al_2O_3 and TiO_2 powders were first mixed for 1 h in a ball mill (QMM/B, Xianyang Jinhong Mechanical Co. Ltd., China) with Al_2O_3 ball and ethanol (the weight ratio for the inorganic powders: Al_2O_3 ball:ethanol was 1:1.5:0.8). Subsequently various organic additives including binders (carboxymethyl cellulose, 2 wt.% with respect to the mixture of Al_2O_3 and TiO₂; all wt.% are relative to the amount

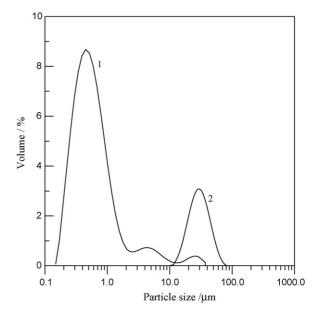


Fig. 1. Particle size distributions of titania (1) and alumina (2) powders.

of inorganic powder), plasticizer (polyvinylalcohol, 10 wt.%), lubricant (glycerine, 7.5 wt.%) and solvent (water, 4.5 wt.%) were slowly added to the powder mixture. The powders plus additives were subsequently purged under vacuum (-0.1 MPa)for 1 h to obtain a homogeneous paste suitable for extrusion. Extrusion was performed at a pressure of 4 MPa using a homemade extruder equipped with a de-gassing device. The extruded tubular green supports (O.D.: 12 mm; I.D.: 8 mm) were dried and sintered in an electrical furnace (SX2-14-17, Wuxi Universal Electrical Machine Works Co., Ltd., China) under air. Four kinds of tubular supports with different compositions are hereafter referred as A100/0 (Al₂O₃/TiO₂ = 100/0, weight ratio), A95/5 (Al₂O₃/TiO₂ = 95/5), A85/15 (Al₂O₃/TiO₂ = 85/15) and A70/30 (Al₂O₃/TiO₂ = 70/30). At the end of each sintering cycle, some supports were taken out from the furnace and air-quenched, while others were cooled down at a rate of 3 °C/min in the furnace to the ambient temperature. The sintered supports with different cooling mode were hereafter designated as A100/0(AQ)-A70/30(AQ) for air-quenched mode and A100/0-A70/30 for cooling down in the furnace to the ambient temperature at a rate of 3 °C/min.

2.3. Characterization of the macroporous supports

The density of sintered supports was determined using Archimedes method with water as immersion medium. Pore size distribution of supports and membranes was measured by a gas bubble pressure method.²¹ Permeability of supports as well as membranes was characterized by pure water flux using cross-flow filtration equipment under various transmembrane pressures (0.05–0.3 MPa) at 25 °C. A plot of permeability (m³ m⁻² h⁻¹) versus transmembrane pressure (bar) is then obtained and the pure water flux (m³ m⁻² h⁻¹ bar⁻¹) of the support/membrane can be calculated from these results. Scanning electron microscope (SEM, JSM-6300, JEOL) was employed

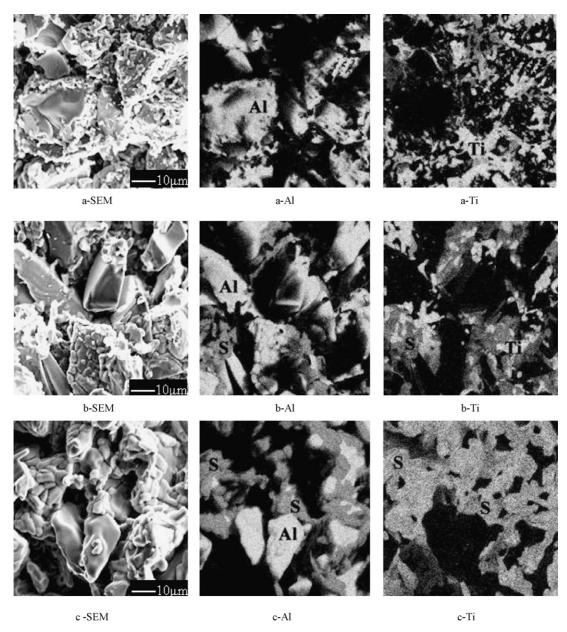


Fig. 2. SEM photos and corresponding micromaps of support A85/15(AQ) sintered at 1300 °C (a-SEM, a-Al, a-Ti), 1400 °C (b-SEM, b-Al, b-Ti) and 1500 °C (c-SEM, c-Al, c-Ti). Al, Al₂O₃; Ti, TiO₂; S, Al₂TiO₅. The analysis area is 80 μ m × 80 μ m.

to observe surfaces of fractured samples as well as the inner surface section of supports and membranes. X-ray micromapping (attached to the SEM, JSM-6300, JEOL) on selected areas ($80 \,\mu\text{m} \times 80 \,\mu\text{m}$) was used to evaluate the Al and Ti elemental distribution of the sintered supports. For each support 3 areas of $80 \,\mu\text{m} \times 80 \,\mu\text{m}$ were analyzed by EDX. The phase composition of sintered supports was probed by X-ray diffraction (XRD, DMAX-RB diffractometer, Rigaku, Japan) with Cu K α radiation. The mechanical strength of tubular supports was determined based on the three point bending method.² A support with 120 mm in length and 12 mm × 8 mm (O.D. × I.D.) in cross section was placed on two stainless-steel jigs 100 mm apart. Samples were bent and fractured by an upper stainlesssteel jig, at a constant loading rate of 550 N/min. The three point bending strength of supports was calculated by the following equation²²:

$$R_f = \frac{8L'}{\pi} \times \frac{P_f(d+2s)}{(d+2s)^4 - d^4} \tag{1}$$

where R_f is the three point bending strength (MPa), P_f is the fracture stress (N), L' is the distance between supporting jigs (mm), d is the inner diameter of tubular support at fracture point (mm), and s is the wall thickness of the support at fracture point (mm). The several strength data obtained from Eq. (1) can be compared regardless of the shape (dimensions) of the tubular supports.

The corrosion resistance of the support was characterized by analyzing the strength of supports after immersing in 1 wt.% NaOH solution (pH value more than 14) and 1 wt.% HNO₃ solu-

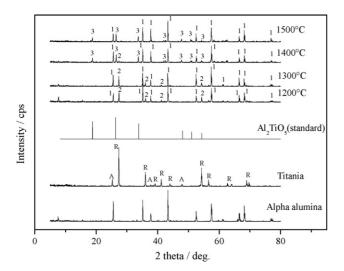


Fig. 3. XRD patterns of starting powders and support A85/15(AQ) sintered at different temperatures (A, anatase; R, rutile; 1, alpha-alumina; 2, rutile; 3, aluminum titanate).

tion (pH value less than 1) for a certain period of time. Sealed PTFE beakers containing the NaOH and HNO₃ solution, respectively were placed into a thermostatic water bath at a constant temperature of $90 \,^{\circ}$ C.

3. Results and discussion

3.1. Microstructure evolution of the Al_2O_3/TiO_2 composite supports as function of sintering temperature

Fig. 2 shows SEM photos of the quenched support A85/15(AQ) sintered at various temperatures. Corresponding to each SEM photo, Al and Ti micromaps reflect the distribution of Al_2O_3 and TiO_2 in the support. It can be seen in Fig. 2a that coarse-grained Al₂O₃ and fine aggregated TiO₂ grains are present after sintering at 1300 °C. As the sintering temperature increased, the amount of fine TiO₂ particles decreased (Fig. 2b). In the corresponding Al and Ti micromaps (Fig. 2b-Al and -Ti), a substance containing both elements Al and Ti ("S" in Fig. 2) is visible. When the sintering temperature reached 1500 °C, the SEM photo in Fig. 2c shows that the particles impinged onto each other and TiO₂ is hardly distinguishable besides alumina. Meanwhile a considerable quantity of the substance containing both elements Al and Ti is present in the corresponding micromaps ("S" in Fig. 2c-Al and -Ti). The XRD patterns (Fig. 3) of support A85/15(AQ) indicated that the substance "S" on the micromaps can be ascribed to aluminum titanate (Al₂TiO₅).

The phase composition of starting powders, together with the support A85/15(AQ) sintered at various temperatures is depicted in Fig. 3. With increasing sintering temperature the phase composition of the support transformed in the following sequence: $Al_2O_3 + TiO_2$ (1200°C and $1300 \circ C$) $\rightarrow Al_2O_3 + TiO_2 + Al_2TiO_5$ $(1400 \,^\circ\text{C}) \rightarrow \text{Al}_2\text{O}_3 + \text{Al}_2\text{TiO}_5$ (1500 $\,^\circ\text{C}$). Fig. 4 shows XRD patterns of sintered and furnace-cooled supports A85/15. If compared with the corresponding XRD patterns of support A85/15(AQ) in Fig. 3, the Al₂O₃/TiO₂ composite support

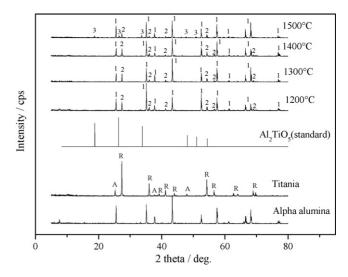


Fig. 4. XRD patterns of starting powders and support A85/15 sintered at different temperatures (A, anatase; R, rutile; 1, alpha-alumina; 2, rutile; 3, aluminum titanate).

cooled down in the furnace at a rate of 3 °C/min always consists of TiO₂ and Al₂O₃ irrespective of the sintering temperature except a small amount of Al₂TiO₅ present in the support sintered at 1500 °C.

It was reported^{18,23} that two concurrent processes occur during sintering of an Al₂O₃/TiO₂ mixture, i.e. densification and a solid-state reaction between Al₂O₃ and TiO₂. At temperatures above 1280 °C, Al₂O₃ reacts with TiO₂ under the formation of Al₂TiO₅, which is also confirmed by the Al₂O₃-TiO₂ phase diagram.²⁴ During cooling to ambient temperature the aluminum titanate decomposes into Al₂O₃ and TiO₂.¹⁵ It should be noted that the formation temperature of Al₂TiO₅ in this paper was observed only around 1400 °C, which is higher than the reported value of 1280–1300 °C.^{15,18,23} This is probably because the particle size of Al₂O₃ ($D_{50} = 30 \,\mu\text{m}$) and TiO₂ ($D_{50} = 0.5 \,\mu\text{m}$) used in this paper was different from the Al₂O₃-TiO₂ powder mixtures (Al₂O₃ ($D_{50} < 0.4 \,\mu\text{m}$)-TiO₂ ($D_{50} = 1.5 \,\mu\text{m}$)²³ and Al₂O₃

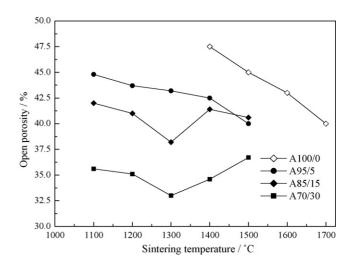


Fig. 5. The variation of porosity of supports as function of sintering temperature (dwell time: 2 h).

Table 2 Average pore size (μ m) of supports with different Al₂O₃/TiO₂ composition sintered at various temperatures.

Sintering temperature (°C)	A95/5	A85/15	A70/30
1100	_	_	0.9
1200	4.7	3.1	1.4
1300	6.4	4.4	2.3
1400	6.5	6.8	4.6
1500	6.1	6.3	6.3

"-" indicates the strength of support is too low to be characterized.

 $(D < 0.4 \,\mu\text{m})$ -TiO₂ $(D_{50} = 13 \,\mu\text{m})^{18}$) studied by Freudenberg and Mocellin. In addition, the higher formation temperature of Al₂TiO₅ reported here can also be ascribed to the low extrusion pressure (4 MPa) compared with a pressure of 250 MPa in the sample fabrication by using an isostatic pressing shaping method.^{18,23} However it can also be the case that the formation of Al₂TiO₅ in support A85/15(AQ) starts at lower temperature (i.e. 1300 °C), but that the amount was very low and it decomposes again into Al₂O₃ and TiO₂ during quenching.

Freudenberg and Mocellin²³ proposed that the formation of Al₂TiO₅ in an Al₂O₃/TiO₂ mixture was the result of Al transport through the TiO₂ layer and reacted with TiO₂. That is, TiO₂ has a double role as a reactant and as rapid Al-transporting medium, which leads to the Al₂TiO₅ formation. The formation of Al₂TiO₅ is very helpful in accelerating the sintering rate of the Al₂O₃/TiO₂ composite support, so sinter necks between alumina particles with size of ~30 μ m could be formed at a relatively low temperature.

3.2. Effect of sintering temperature on support properties

The open porosity of the supports A100/0–A70/30 sintered at various temperatures and furnace-cooled is displayed in Fig. 5. Because of the extremely low mechanical strength of the support A100/0 sintered below 1400 °C, only the porosity of this support sintered above 1400 °C is displayed. It is obvious that doping titania into alumina support results in a better sintering behaviour as evidenced by the decrease in support A95/5, it can be seen that the open porosity decreased with increasing sintering temperature. However, an irregular result for the variation of the open porosity is apparent for supports A85/15 and A70/30.

As was mentioned in Section 3.1, the heat treatment of an Al_2O_3/TiO_2 composite support consists of two concurrent pro-

Table 3

Three point bending strength (MPa) of supports with different Al_2O_3/TiO_2 composition sintered at various temperatures.

Sintering temperature (°C)	A95/5	A85/15	A70/30
1100	_	7.3	23.1
1200	-	18.1	40.7
1300	13.1	34.1	49.4
1400	18.2	32.7	13.2
1500	26.3	26.2	1.6

"-" indicates the strength of support is too low to be determined.

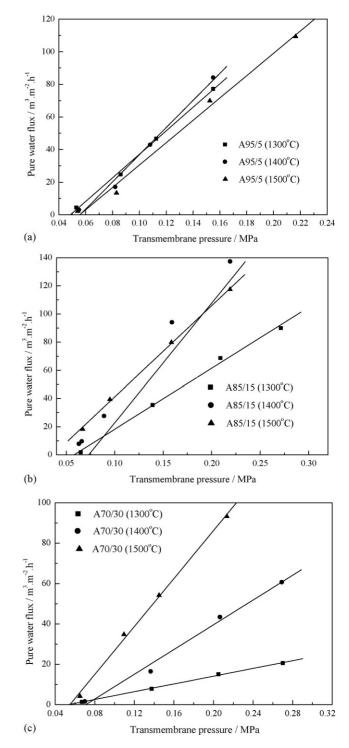


Fig. 6. Pure water flux of supports sintered at various temperatures versus transmembrane pressure: (a) A95/5, (b) A85/15, and (c) A70/30 (dwell time: 2 h).

cesses of densification and reaction. The theoretical density of β -Al₂TiO₅, α -Al₂O₃ and rutile is 3.70 g/cm³, 3.90 g/cm³ and 4.25 g/cm³, respectively. Therefore, the reaction between Al₂O₃ and TiO₂ is accompanied by an 11% molar volume increase.²³ The open porosity of the support (A95/5–A70/30) decreased regardless of the composition of Al₂O₃/TiO₂ because only densification took place in the supports below 1300 °C.

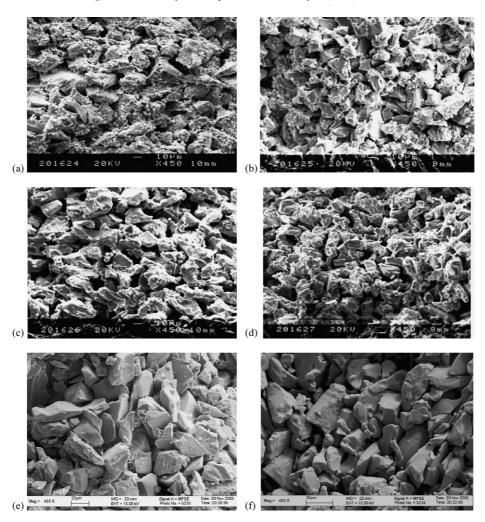


Fig. 7. SEM sectional photos of the support A85/15 sintered at various temperatures of (a) $1200 \,^{\circ}$ C, (b) $1300 \,^{\circ}$ C, (c) $1400 \,^{\circ}$ C, (d) $1500 \,^{\circ}$ C, as compared with the support A100/0 sintered at (e) $1600 \,^{\circ}$ C and (f) $1700 \,^{\circ}$ C, respectively.

When the sintering temperature is higher than $1300 \,^{\circ}$ C, the porosity of Al₂O₃/TiO₂ composite support will decrease further because of the densification process. However, the simultaneously occurring volume expansion caused by the Al₂O₃/TiO₂ reaction hampers this densification. Therefore, the amount of Al₂TiO₅ formed in the support plays a key role in the variation of the open porosity. If the porosity reduction caused by densification does not match with the volume expansion due to aluminum titanate formation, the open porosity of the support would decrease (in the case of support A95/5) or increase (in the case of support A70/30) as the sintering temperature increased. If porosity reduction matches the volume expansion due to the formation of the aluminum titanate, the porosity of the support does not vary obviously with increasing sintering temperature (in the case of support A85/15 sintered at 1400 °C and 1500 °C).

In Table 2 average pore sizes of supports A95/5–A70/30 sintered at various temperatures are given. Except for the support A95/5 and A85/15 sintered at 1500 °C, it can be seen that the pore size of support increases as the sintering temperature increases irrespective of the TiO₂ doping amount. Fine-grained TiO₂ occupying the space formed by the packing of coarse-grained Al₂O₃ should be responsible for the smaller pore size of support sintered at low temperatures ($1200 \,^{\circ}$ C or lower). As the sintering temperature increases, the disappearance of TiO₂ and formation of Al₂TiO₅ accordingly increase the distance between Al₂O₃ particles and result in an increase in pore size.²⁵ It should be noted that the pore size of the supports A95/5 and A85/15 sintered at 1500 $^{\circ}$ C is (slightly) smaller than that of those supports sintered at 1400 $^{\circ}$ C. This result can be attributed to the densification of a support consisting of Al₂O₃ and Al₂TiO₅ after the completion of reaction of Al₂O₃–TiO₂ composite at 1400 $^{\circ}$ C.

Fig. 6 gives the pure water flux of the support as function of transmembrane pressure. By comparison with Fig. 5 and Table 2, it is obvious that the pore size has a stronger impact on pure water flux of support than porosity.

Table 3 shows three point bending strengths of supports sintered at different temperatures. The strength of support A95/5 increased with sintering temperature. Compared with support A95/5, the mechanical strength of support A85/15 sintered below 1400 °C exhibits the same tendency. It is worthwhile to note that a higher temperature of 1500 °C leads to a lower mechanical strength for the support A85/15. With respect to the support A70/30, a maximum in strength of 49.4 MPa is obtained at a sintering temperature of 1300 °C. The higher

sintering temperature above 1300 °C impairs the strength of support A70/30 greatly. The same phenomenon observed in the literature^{18,19} indicates that the sintering temperature of Al₂O₃/TiO₂ system must not be higher than 1275 °C because the appearance of Al₂TiO₅ would impair the mechanical strength of the Al₂O₃/TiO₂ composite. Our results confirm the previous observations²³ and find that the amount of Al₂TiO₅ plays an important role in determining the mechanical strength of Al₂O₃/TiO₂ composite supports. The more the Al₂TiO₅ formed in the Al₂O₃/TiO₂ composite, the lower the strength of the supports. Therefore, to obtain a macroporous membrane supports exhibit both high mechanical strength and high permeability, the sintering temperature for support A85/15 should not be higher than 1400 °C.

Fig. 7 shows the microstructure evolution for support A85/15 at different sintering temperatures, together with SEM images for support A100/0 obtained after sintering at 1600 °C and $1700 \,^{\circ}$ C. Support A85/15 shows that the fine particles (TiO₂) and coarse-grained Al₂O₃ are aggregated at a sintering temperature of 1200 °C. The fine TiO₂ particles gradually disappeared and most of the particles impinged onto each other as the sintering temperature elevated. With respect to the support sintered above 1400 °C, TiO₂ particles can hardly be discriminated and sinter necks between alumina particles are visible. However, as far as the pure alumina support is concerned, there is only a slight difference between the microstructure of supports sintered at 1600 °C and 1700 °C (Fig. 7e and f). Only a packing of alumina particles is visible and no obvious sinter necks could be observed even if it is sintered at 1700 °C, as evidenced by the low mechanical strength of this support (<10 MPa). With respect to support A85/15, it is worthwhile to note that the sinter neck area increased when sintering at 1500 °C. However, the lower strength of this support than that of 1400 °C sintered one confirmed the negative influence of the formation of Al₂TiO₅ on the mechanical strength of the Al₂O₃/TiO₂ composite supports.

Chemical stability tests for A85/15 (sintered at 1400 °C for 2 h) were performed by immersing the support into 1 wt.% NaOH solution (90 °C) and 1 wt.% HNO₃ solution (90 °C), respectively for a certain period of time. The results shown in Fig. 8a and b indicate that the support exhibits an excellent chemical stability towards NaOH and HNO₃ solutions (90 °C) even after a corrosion test for 720 h.

According to the above investigations, the macroporous composite support A85/15 sintered at 1400 °C for 2 h with integrated properties of permeability, mechanical and chemical stability was chosen for microfiltration (MF) membranes preparation. A well-dispersed and stable submicron alumina suspension was synthesized and subsequently coated onto the support A85/15 via a dip-coating method and calcined at 1300 °C afterwards. The mean pore size of the MF membrane coated on the composite support A85/15 was 0.7 μ m and more than 90% pores were in the range of 0.4–2 μ m, as evidenced by Fig. 9. The pure water flux of the MF layer was 10 m³ m⁻² h⁻¹ bar⁻¹. Microstructures of the surface and the fracture section of the membrane shown in Fig. 10a and b confirmed a crack-free membrane layer with a thickness in the range of 30–40 μ m.

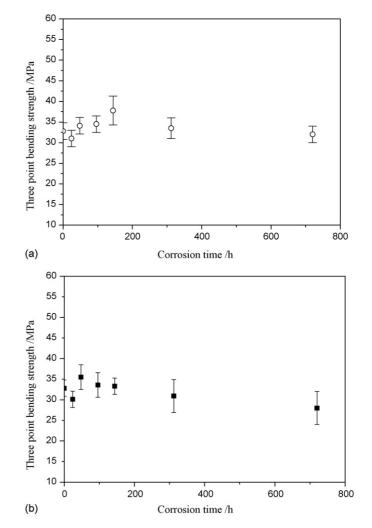


Fig. 8. Variation of three point bending strength of support A85/15 as function of time immersed in corrosive liquids (at 90 $^{\circ}$ C): (a) 1 wt.% NaOH solution and (b) 1 wt.% HNO₃ solution.

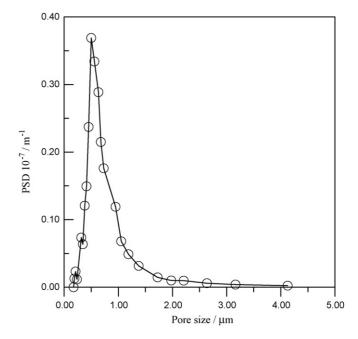
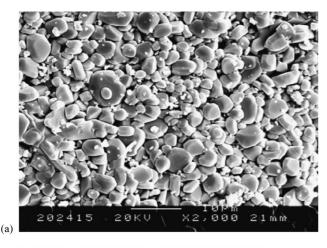


Fig. 9. Pore size distribution of the MF membrane coated on supports A85/15.



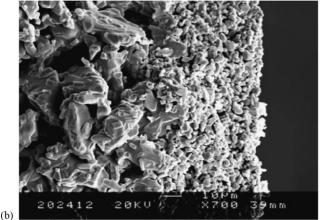


Fig. 10. SEM photos of surface (a) and fractured (b) section of the MF membrane.

4. Conclusions

- (1) A tubular macroporous ceramic support consisting of alumina and titania showing sufficient permeability and high mechanical strength was prepared by extrusion and subsequent sintering at 1400 °C for 2 h. This low sintering temperature can largely reduce the cost of the macroporous membrane support.
- (2) The formation mechanism of the support was studied in detail by SEM, EDX micromapping and XRD, which confirmed that two concurrent processes occurred during the sintering process of these Al₂O₃/TiO₂ composite support, i.e. densification and solid-state reaction, The formation of Al₂TiO₅ depends on sintering temperature and has a strong impact on the properties of the Al₂O₃/TiO₂ composite support, especially with regard to the mechanical strength.
- (3) The effect of the sintering temperature on the properties of Al_2O_3/TiO_2 composite support was investigated. The results show that the support with a composition of 85 wt.% $Al_2O_3/15$ wt.% TiO_2 sintered at 1400 °C for 2 h exhibits both high permeability (with a pure water flux of 45 m³ m⁻² h⁻¹ bar⁻¹) and sufficient mechanical strength (32.7 MPa), together with excellent corrosive resistance towards hot NaOH and HNO₃ solutions (1 wt.%, 90 °C).

(4) A crack-free MF membrane with a pore size of $\sim 0.7 \,\mu m$ was successfully prepared via dip-coating method by using the above-mentioned Al₂O₃/TiO₂ composite support.

Acknowledgements

The author would like to thank the financial support of the National Basic Research Program (973 Program, Code: 2009CB623400) and the National Natural Science Foundation of China (20906047). This work was also sponsored by State Key Laboratory of Chemical Engineering (No. SKL-ChE-09A01) and the Scientific Research Foundation for the Returned Overseas Chinese Scholars.

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