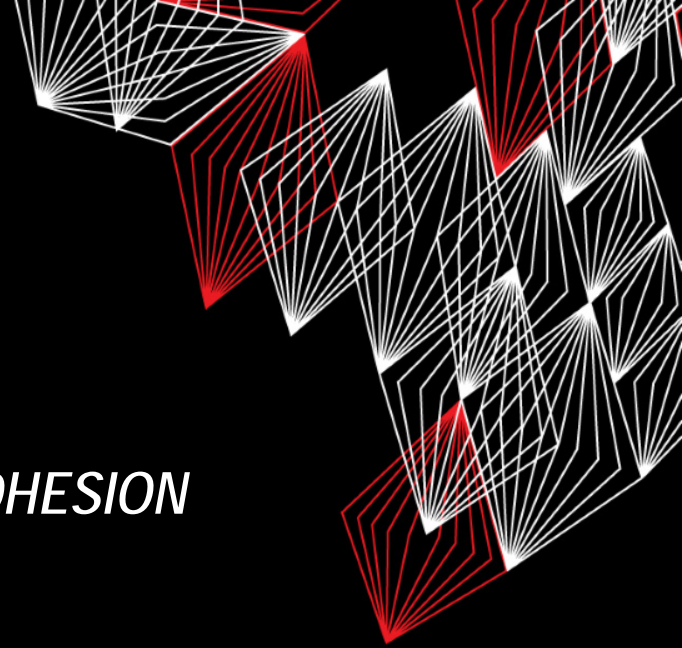


UNIVERSITY OF TWENTE.

3 July 2018,  
Nuremberg, Germany



*MIMICKING THE NATURE: HOOK-AND-LOOP ADHESION  
SYSTEMS FOR ELASTOMERS*

RAFAŁ ANYSZKA<sup>1,2</sup>), WILMA DIERKES<sup>1</sup>), ANKE BLUME<sup>1</sup>),  
DARIUSZ M. BIELINSKI<sup>2</sup>), ESSI SARLIN<sup>3</sup>)

<sup>1</sup>) Chair of Elastomer Technology and Engineering, University of Twente, The Netherlands

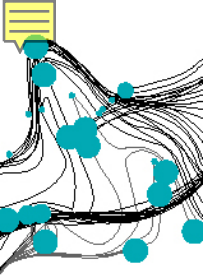
<sup>2</sup>) Institute of Polymer and Dye Technology, Lodz University of Technology, Poland

<sup>3</sup>) Department of Materials Science, Tampere University of Technology, Finland



**DKT 2018**  
Deutsche Kautschuk-Tagung

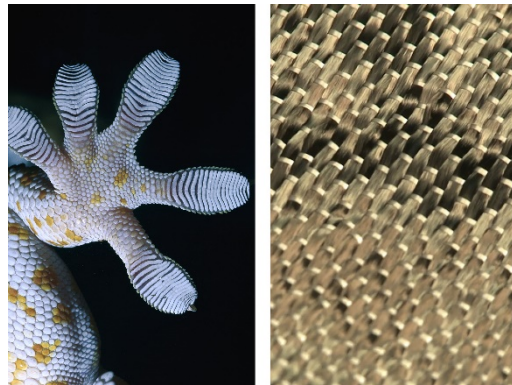




# ORIGIN OF CONCEPTS

WHERE DO IDEAS COME FROM? – BIOMIMICRY: CASE STUDY

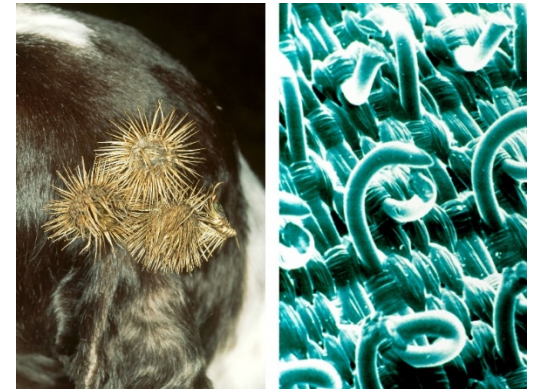
## Gecko Feet Adhesives



## Shark skin



## Velcro

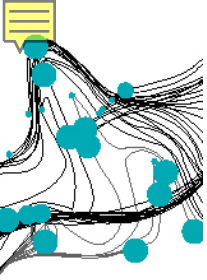


## Annoying feature of weeds



## Useful technical solutions





# INTRODUCTION

## CHARACTERISTICS OF MICROSCOPIC VS. MOLECULAR VELCRO SYSTEMS



### Microscopic Velcro system

- Superior fatigue resistance
- Superior reconnectability performance
- Good mechanical properties
- Good ageing resistance
- Stiff hooks, elastic loops
- Hooks and loops materials chemical compatibility – not relevant
- Molecular mobility – not relevant

VS.

### Molecular Velcro system

- Superior fatigue resistance
- Superior reconnectability performance
- Good mechanical properties
- Good ageing resistance
- Stiff/elastic hooks, stiff/elastic loops
- Hooks and loops materials chemical compatibility – very relevant (mutual solubility/miscibility)
- Molecular mobility – very relevant



# INTRODUCTION

## FOCUS ON PHYSICAL INTERACTIONS

### Chemical interactions:

- Covalent bonds
- Ionic bonds
- Coordinate bonds

### Strong physical interactions:

- Hydrogen bonds
- Ion/dipole and ion-induced/dipole interactions
- Dipole/dipole interactions

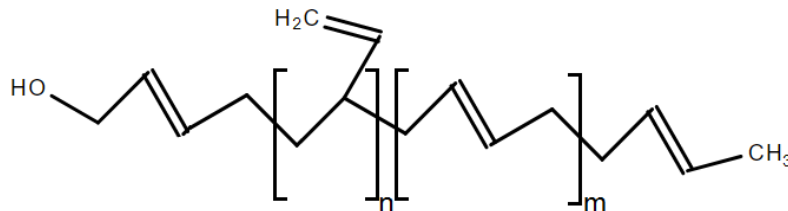


### Physical interactions:

- Dispersion interactions
- Steric hindrance
- Macromolecular entanglements
- Chemical affinity (miscibility/solubility)



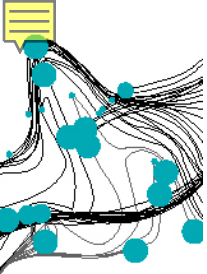
**Velcro-like approach** – grafting of relatively large molecules onto the silica surface of good chemical affinity to the rubber, enhancing interactions *via* physical entanglements and steric hindrance.



Molecular weight of the o-BR: 4691 g/mol  
Length of straightened molecule: ~30 nm  
Number of vinyl mers: ~60 per molecule

*Telechelic mono-hydroxy polybutadiene oligomer (o-BR) was used as a backbone for the modifier.*

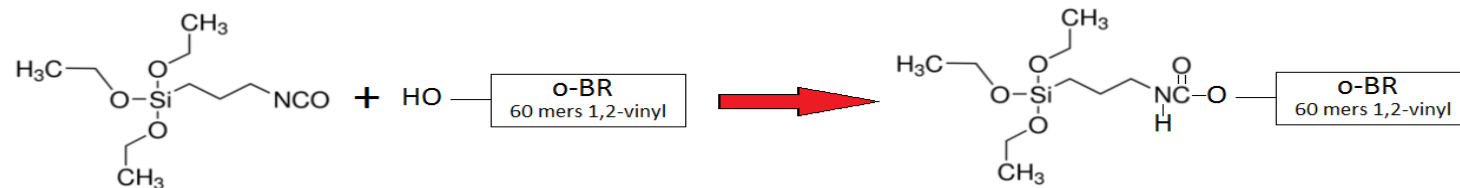




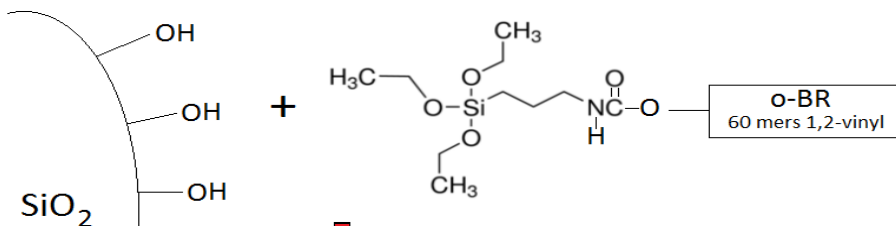
# INTRODUCTION

## SCHEME OF SILICA-SURFACE MODIFICATION

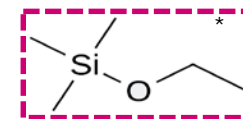
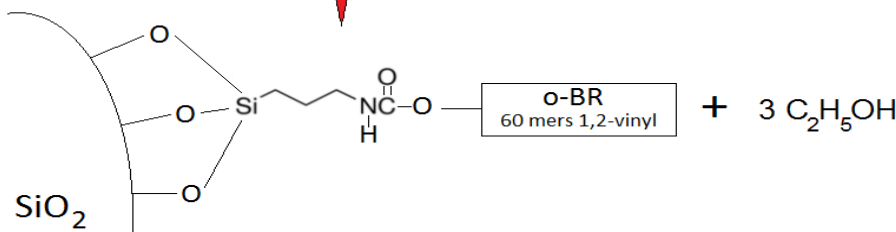
### 1 – Attaching the isocyanate silane molecule to the telechelic o-BR chain



### 2 – Grafting on the silica surface



### 3 - Additional treatment with trimethylethoxysilane in order to cover residual, reactive silanol groups

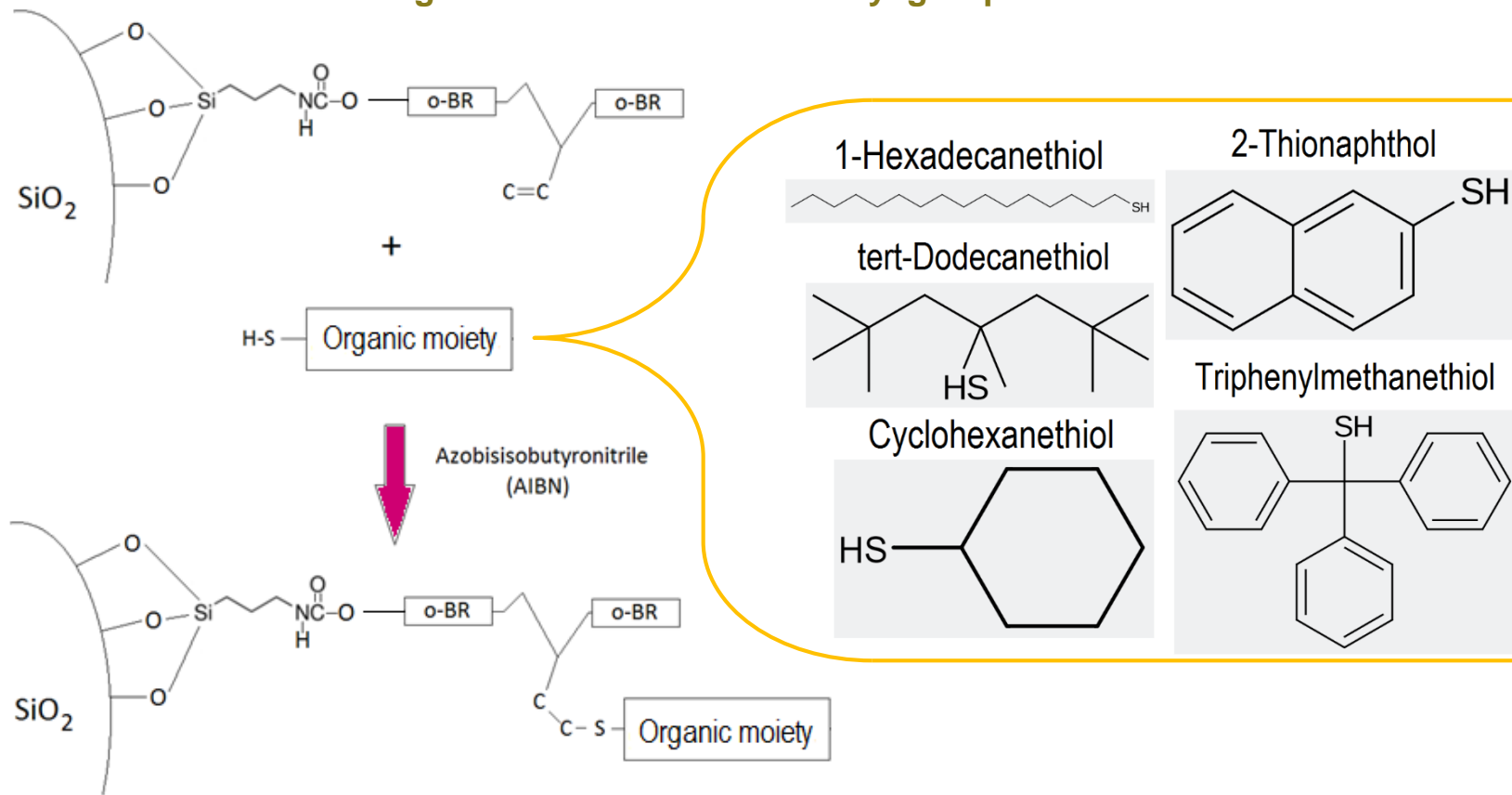




# INTRODUCTION

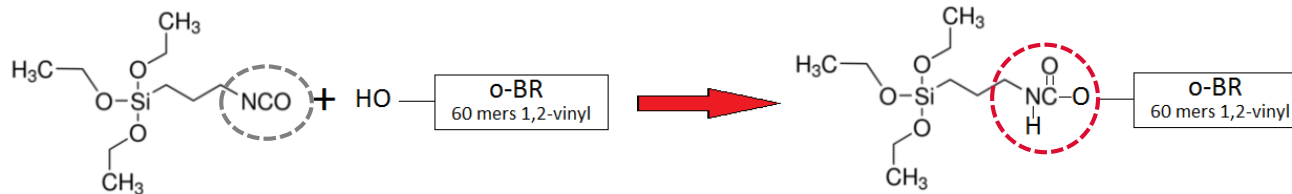
## SCHEME OF SILICA-SURFACE MODIFICATION

### 4 – Attaching thioles moieties to the vinyl groups of o-BR

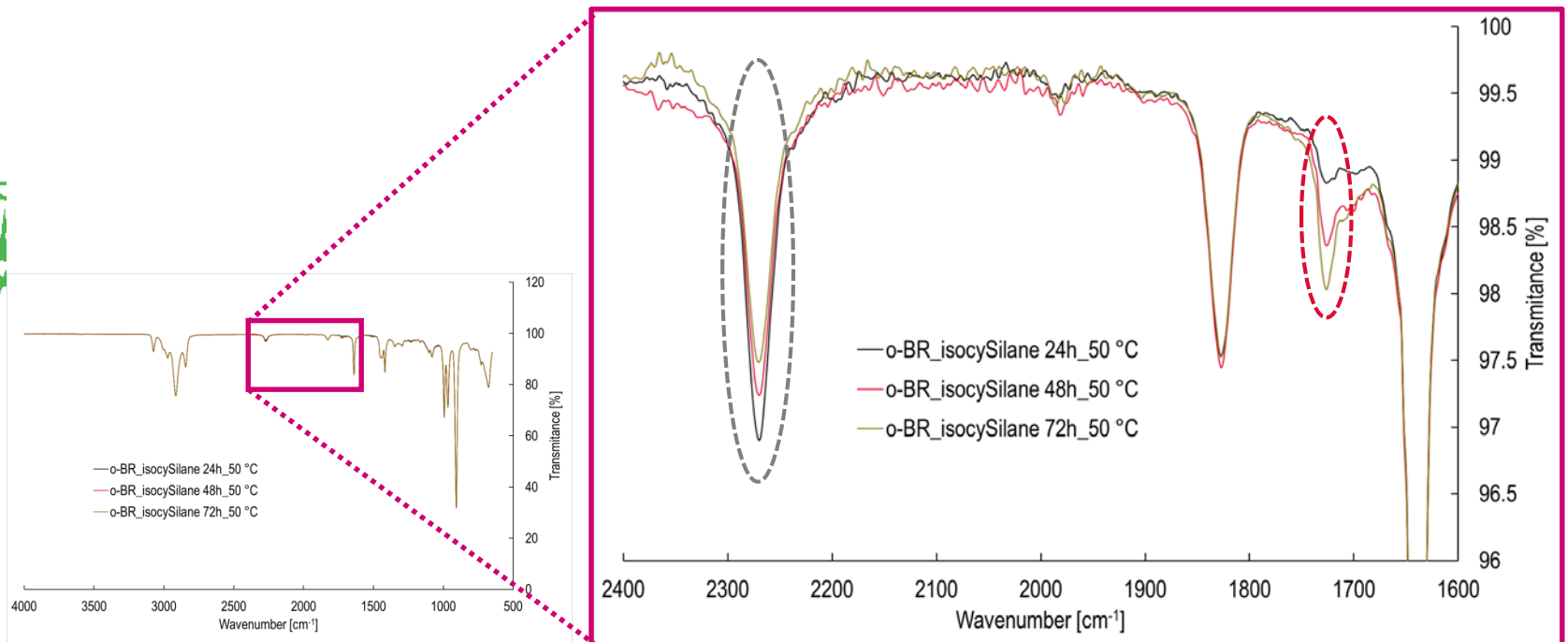


# SYNTHESIS OF OLIGOMER-BACKBONE

## REACTION PROGRESS TRACKING BY FTIR

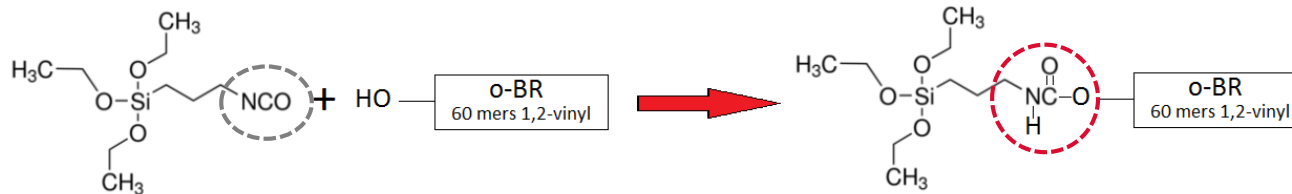


Isocyanate silane – 0.527 g + o-BR – 10 g: mixed together and heated at 50 °C for 24, 48 and 72 hours.

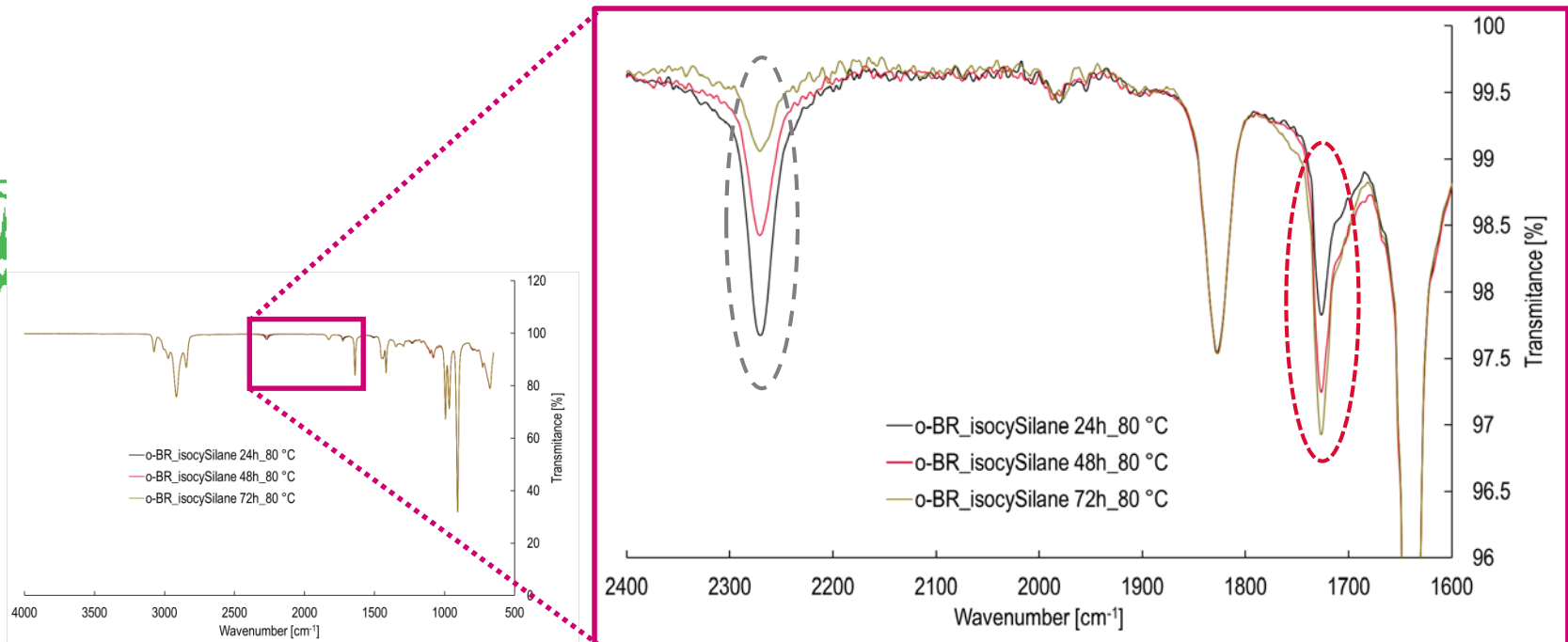


# SYNTHESIS OF OLIGOMER-BACKBONE

## REACTION PROGRESS TRACKING BY FTIR



Isocyanate silane – 0.527 g + o-BR – 10 g: mixed together and heated at 80 °C for 24, 48 and 72 hours.





# SYNTHESIS OF OLIGOMER-BACKBONE

## REACTION PROGRESS TRACKING BY FTIR

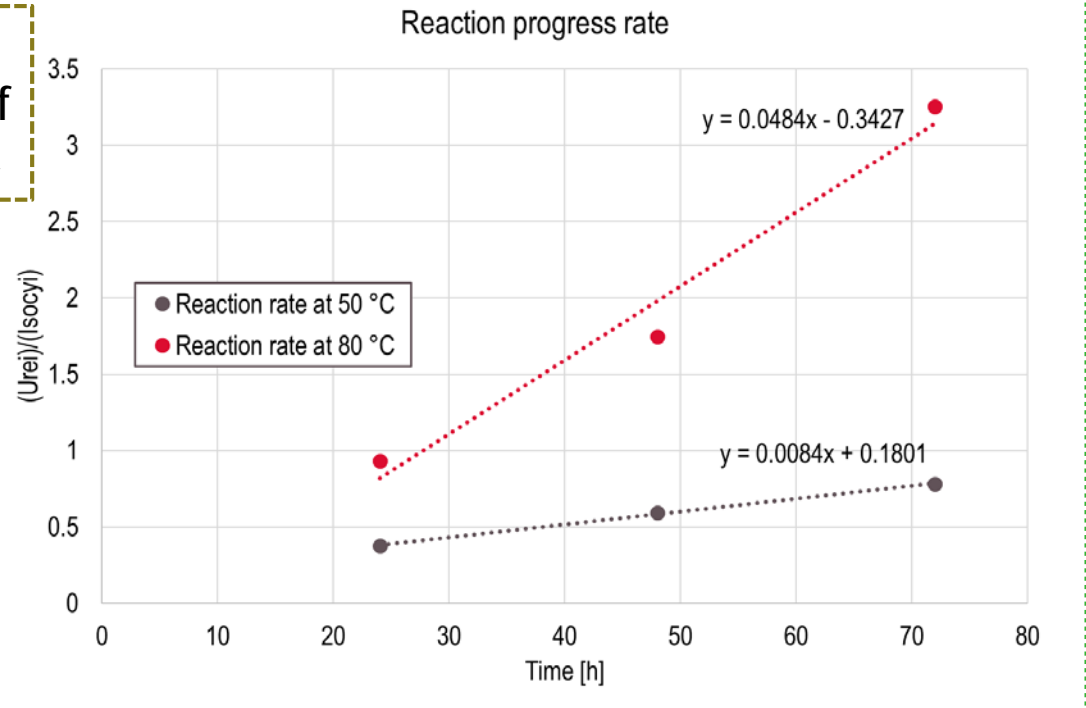
Reaction rate at 50 °C

Time [h]	Isocyanate group intensity ( $I_{\text{Isoc}_i}$ )	Urethane group intensity ( $I_{\text{Ure}_i}$ )	$(I_{\text{Ure}_i}) / (I_{\text{Isoc}_i})$
24	3.0999	1.1704	0.3775
48	2.7610	1.6400	0.5940
72	2.5133	1.9654	0.7820

Reaction rate at 80 °C

Time [h]	Isocyanate group intensity ( $I_{\text{Isoc}_i}$ )	Urethane group intensity ( $I_{\text{Ure}_i}$ )	$(I_{\text{Ure}_i}) / (I_{\text{Isoc}_i})$
24	2.3247	2.1735	0.9350
48	1.5761	2.7515	1.7458
72	0.9432	3.0720	3.2569

The reaction rate increases significantly with the increase of temperature from 50 °C to 80 °C

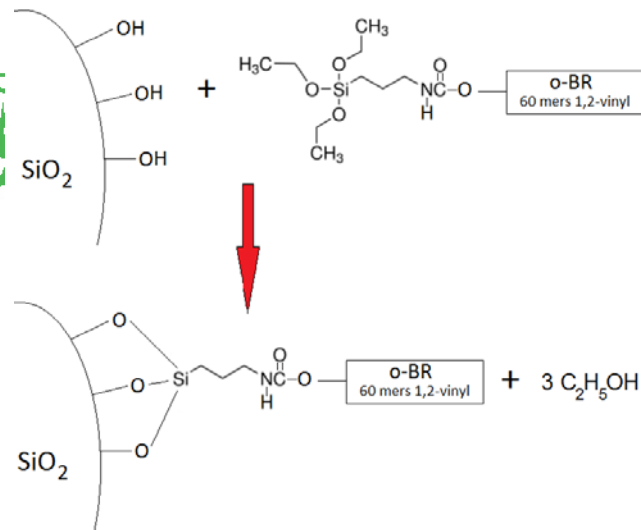


# OLIGOMER-BACKBONE GRAFTING ON SILICA SURFACE

## PROCEDURE CHARACTERISTICS

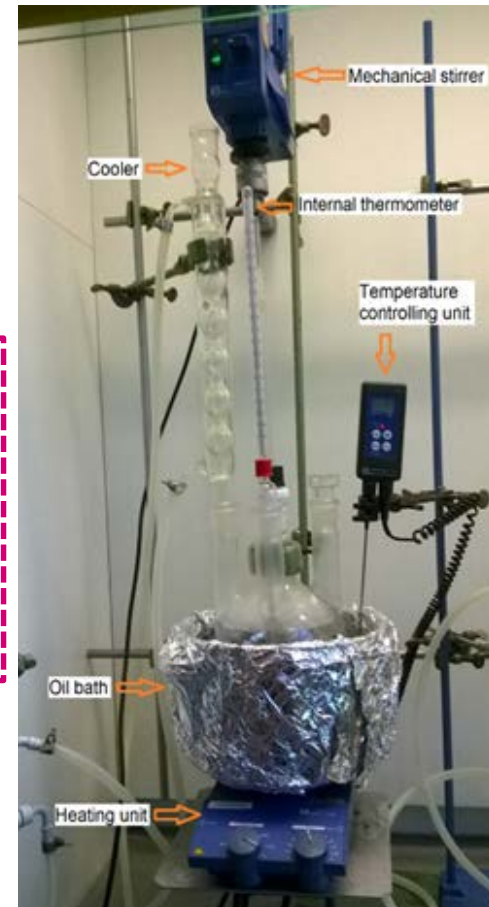
### Composition of the samples

Sample description	Weight ratio	Precipitated silica (MP)	IsocySilane/o-BR
Silica + 20 silane_o-BR_extr20h	1/5	100 g	20 g
Silica + 50 silane_o-BR_extr20h	1/2	100 g	50 g
Silica + 100 silane_o-BR_extr20h	1/1	50 g	50 g



### Procedure:

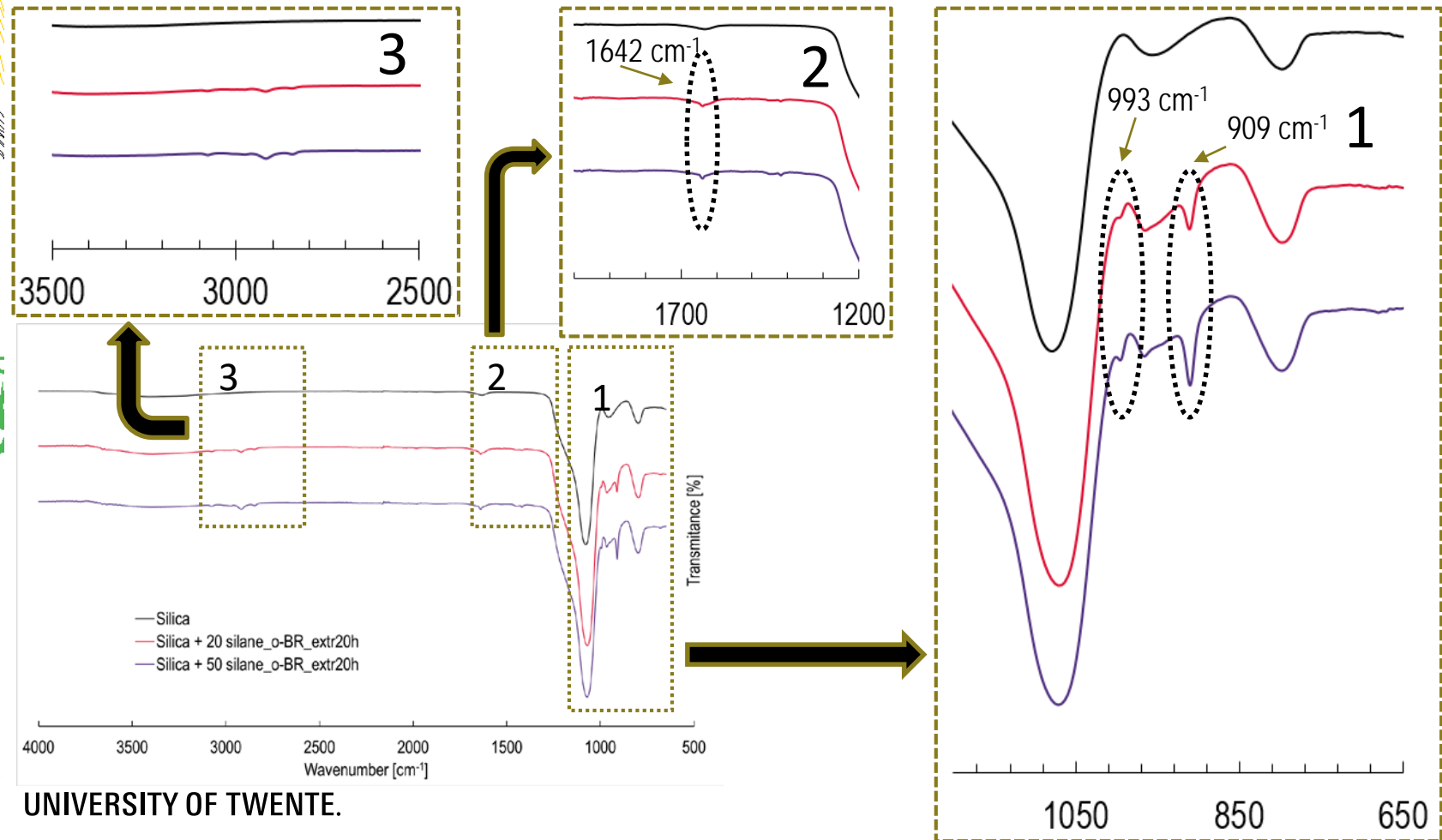
- Duration - 24 hours
- Air atmosphere
- Temperature - 100 °C
- Mechanical stirring – 150 rpm
- Extraction in toluene after the reaction – 20 hours



# OLIGOMER-BACKBONE GRAFTING ON SILICA SURFACE

## GRAFTING RESULTS ANALYSED BY FTIR

FTIR spectra of silica modified with various amounts of the oligomer-backbone (*indicated bands from unsaturated groups*)

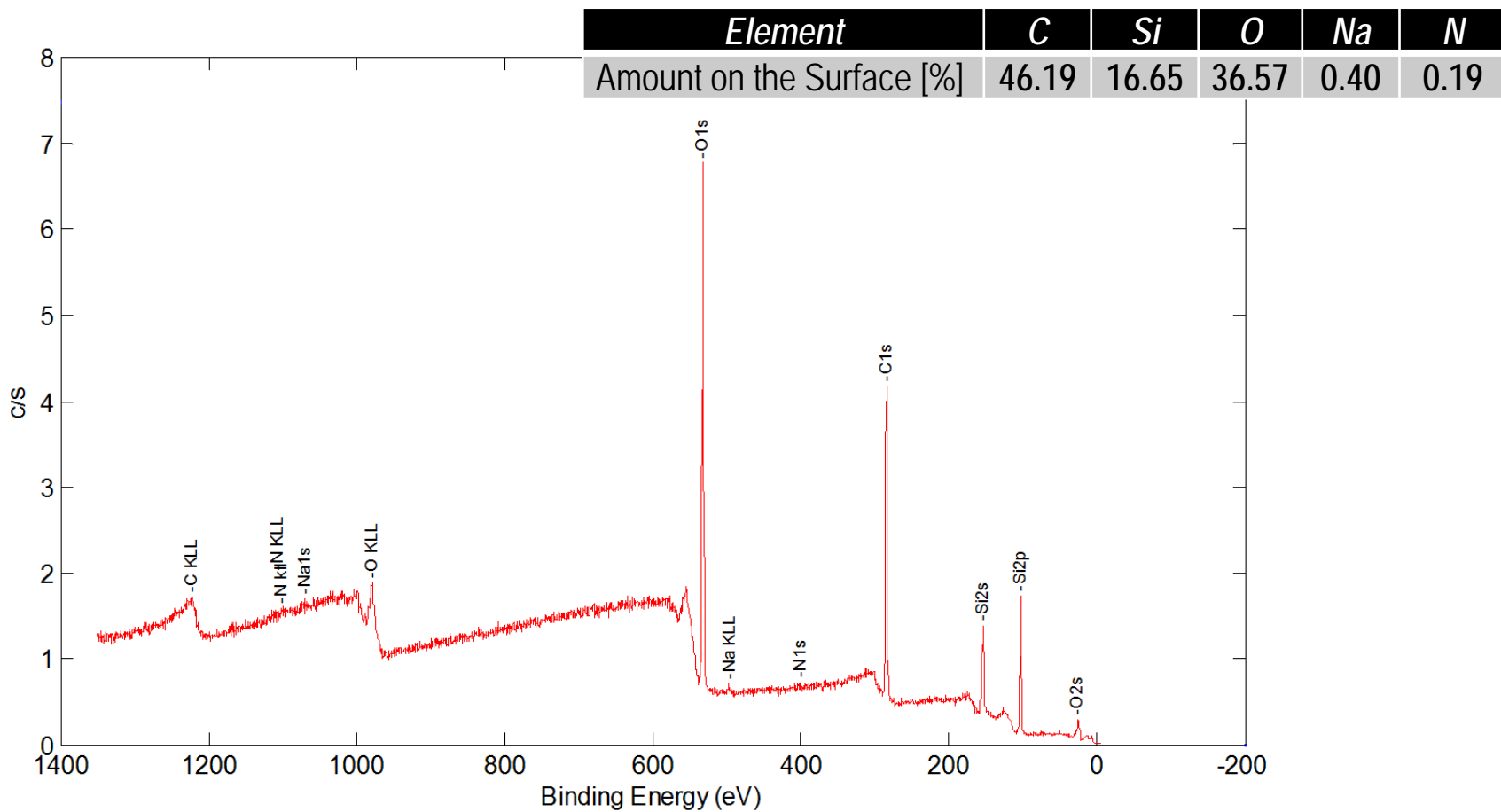




# OLIGOMER-BACKBONE GRAFTING ON SILICA SURFACE

## GRAFTING RESULTS ANALYSED BY XPS

*XPS analysis of the silica sample modified with 50 parts of the o-BR per 100 parts of the silica*



# OLIGOMER-BACKBONE GRAFTING ON SILICA SURFACE

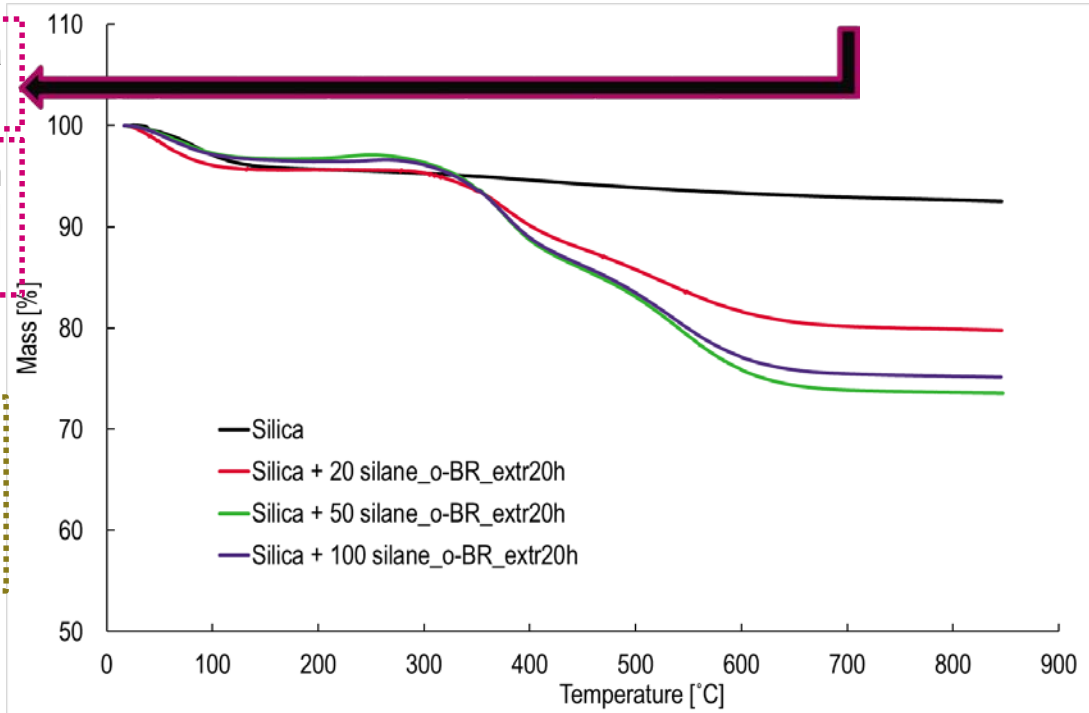
## GRAFTING RESULTS ANALYSED BY TGA

Sample description	Summary mass loss [%]	Mass loss of organic component [%]	Density of grafted molecules [1/nm <sup>2</sup> ]
Silica	7.5	-	-
Silica + 20 silane_o-BR_extr20h	20.2	15.8	6.37
Silica + 50 silane_o-BR_extr20h	26.4	22.9	4.07
Silica + 100 silane_o-BR_extr20h	24.8	21.3	4.46

Average silanol group concentration on silica surface according to literature is ~5 -OH/nm<sup>2</sup>

Single o-BR molecule terminated with the silane can react with up to 3 silanol groups (ca. 10-15 % of all -OH groups)

Additional treatment is required to cover the residual, reactive -OH groups



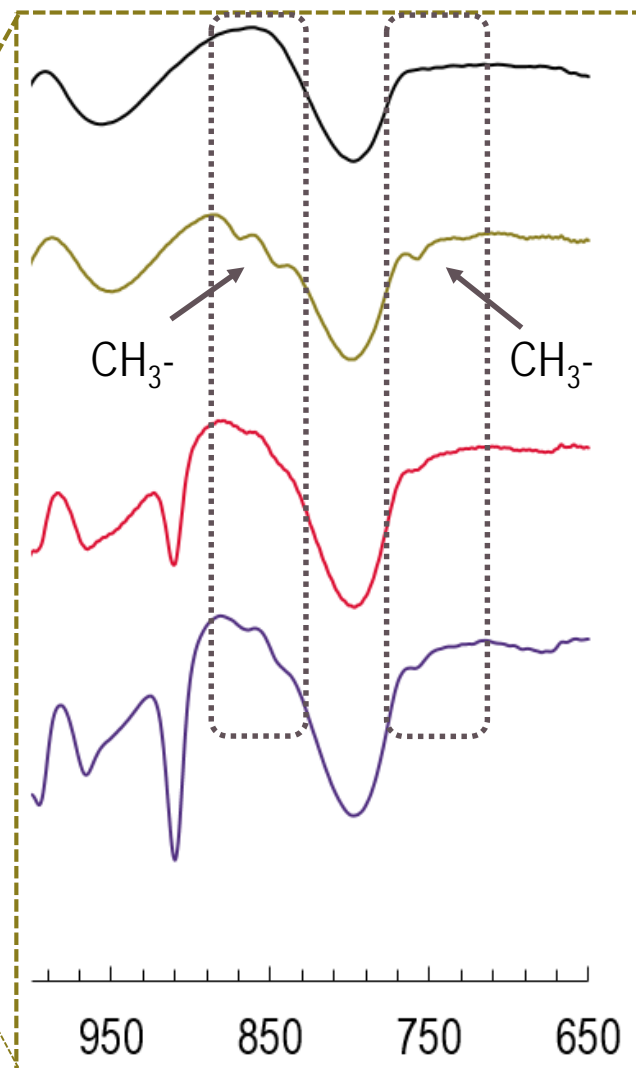
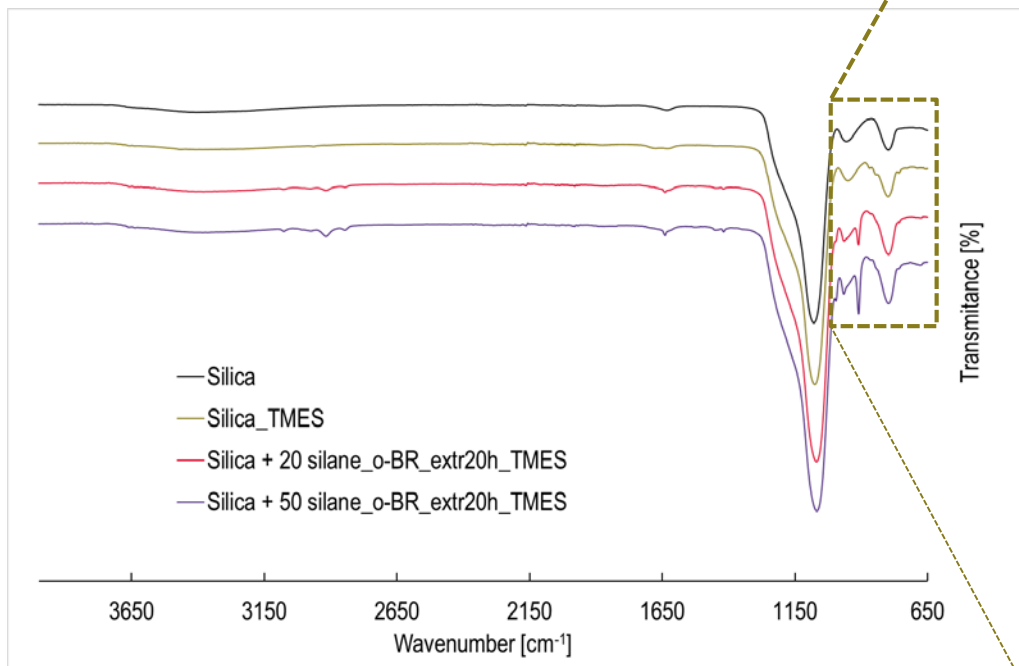
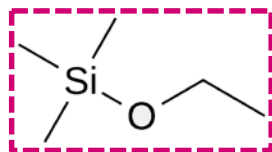
UNIVERSITY OF TWENTE.

# ADDITIONAL SILANIZATION

## SILANIZATION RESULTS ANALYSED BY FTIR

Additional treatment with trimethylethoxysilane in order to cover residual, reactive silanol groups.

- Suspension in toluene
- Temperature: 70 °C
- Duration: 24 hours

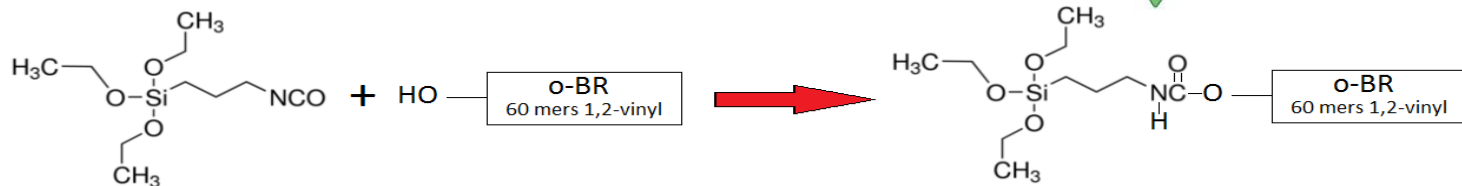




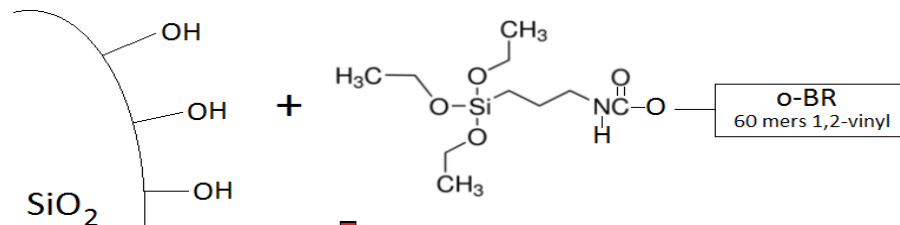
# PROGRESS OF THE MODIFICATION

## SCHEME OF SILICA-SURFACE MODIFICATION

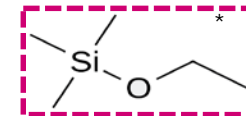
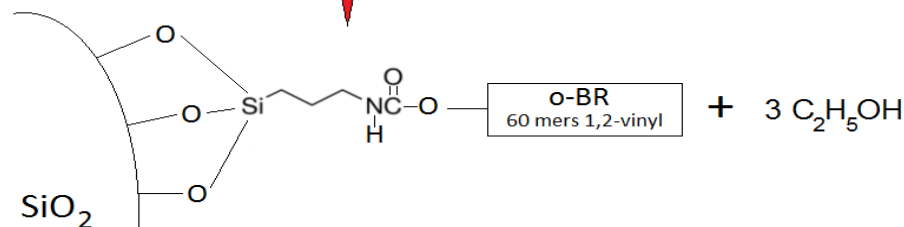
### 1 – Attaching the silane molecule to the telechelic o-BR chain

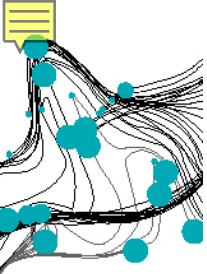


### 2 – Grafting on the silica surface



### 3 - Additional treatment with trimethylethoxysilane in order to cover residual, reactive silanol groups

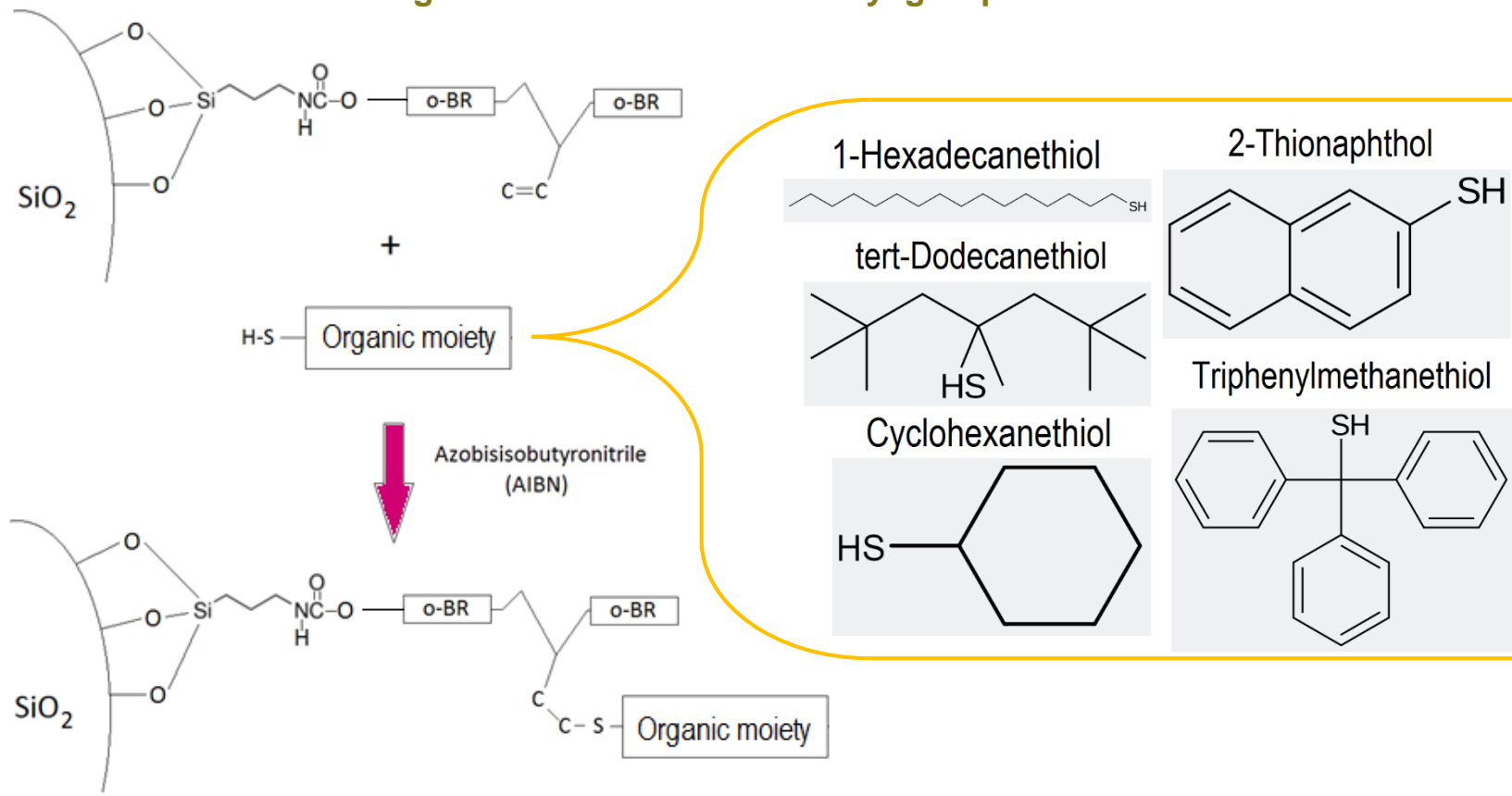




# PROGRESS OF THE MODIFICATION

## SCHEME OF SILICA-SURFACE MODIFICATION

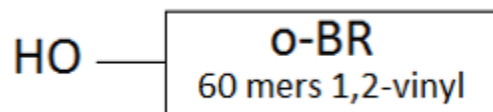
### 4 – Attaching thioles moieties to the vinyl groups of o-BR





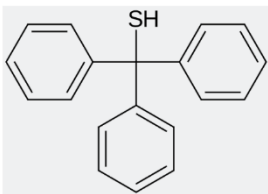
# BRANCHING OF OLIGOMER-BACKBONE

## TELECHELIC BUTADIENE OLIGOMER REACTIONS WITH THE THIOLES

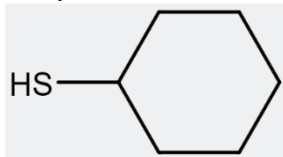


+

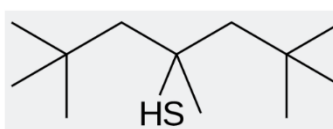
Triphenylmethanethiol



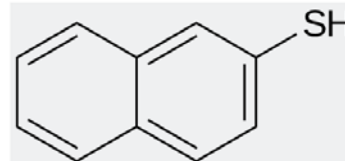
Cyclohexanethiol



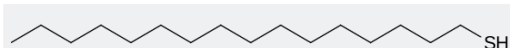
tert-Dodecanethiol



2-Thionaphthol



1-Hexadecanethiol



### Procedure parameters:

Duration – 1 hour

Temperature – 65-70 °C (constant growth)

Mechanical stirring – 150 rpm

Atmosphere – nitrogen

Evaporation – 24 hours, 70 °C

Degassing - 24 hours, 70 °C (vacuum)

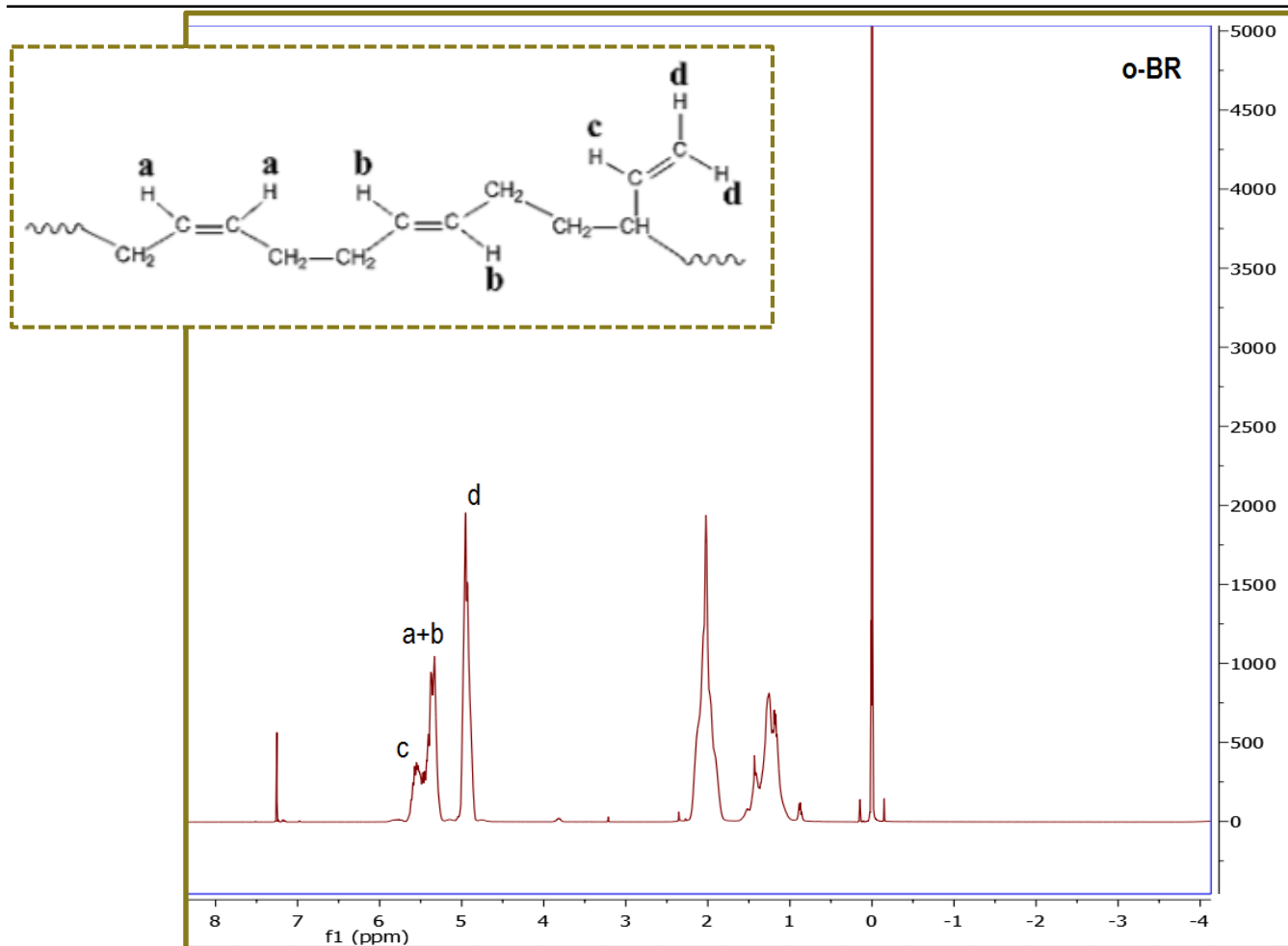
After evaporation

After degassing  
in vacuum oven



# BRANCHING OF OLIGOMER-BACKBONE

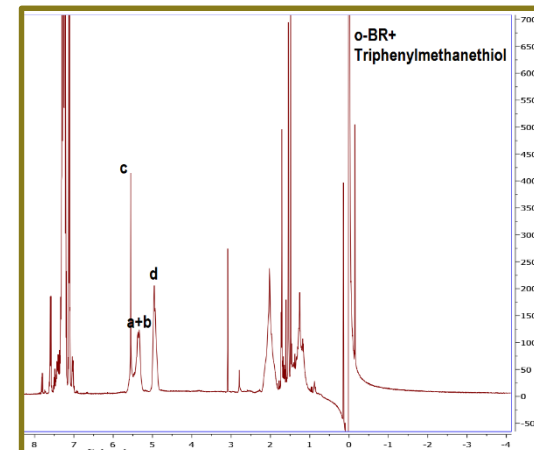
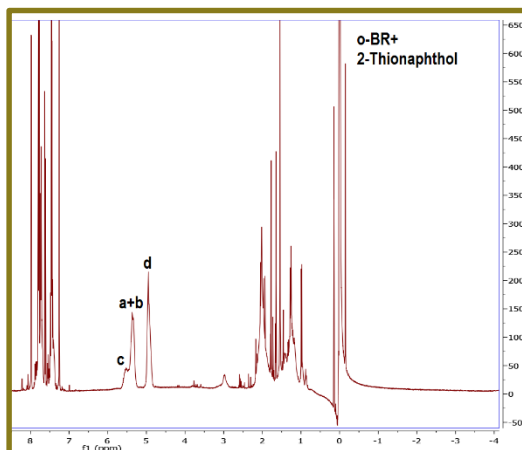
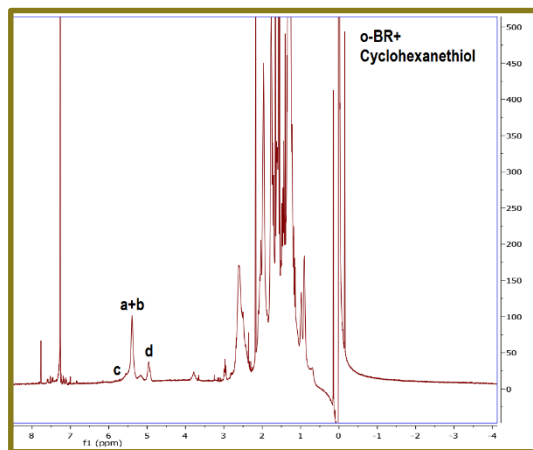
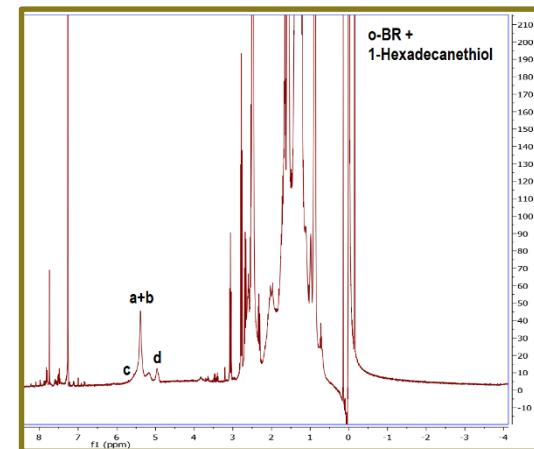
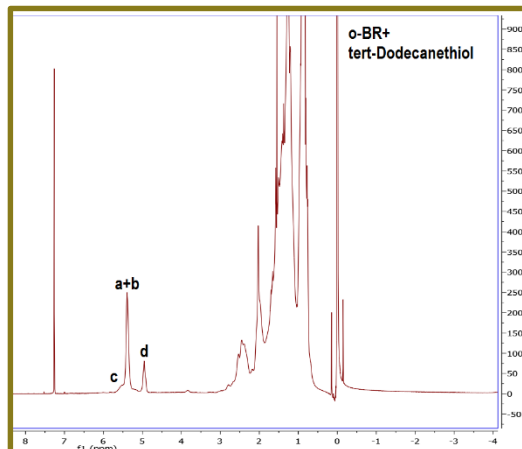
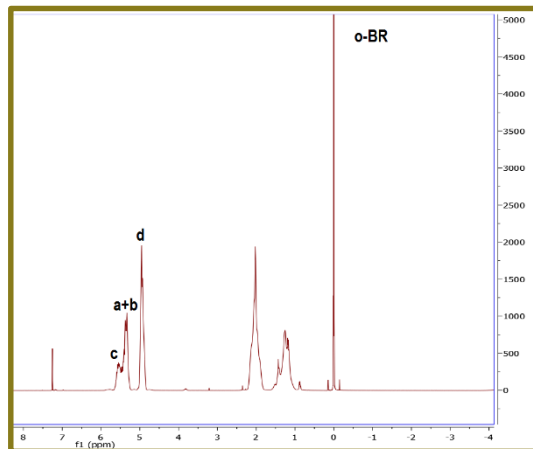
## OLIGOMER BRANCHING RESULTS ANALYSED BY LS-NMR





# BRANCHING OF OLIGOMER-BACKBONE

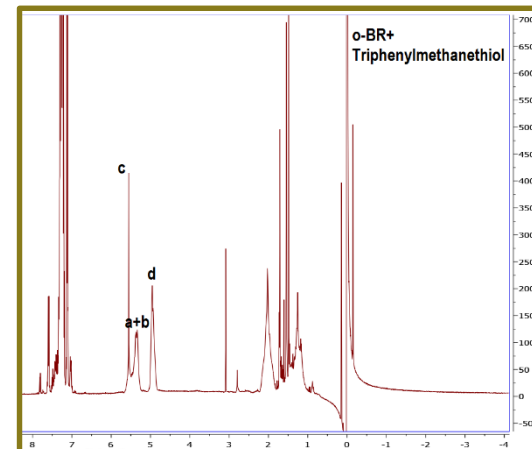
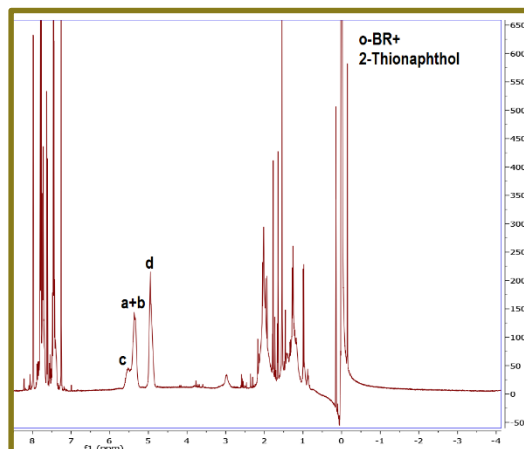
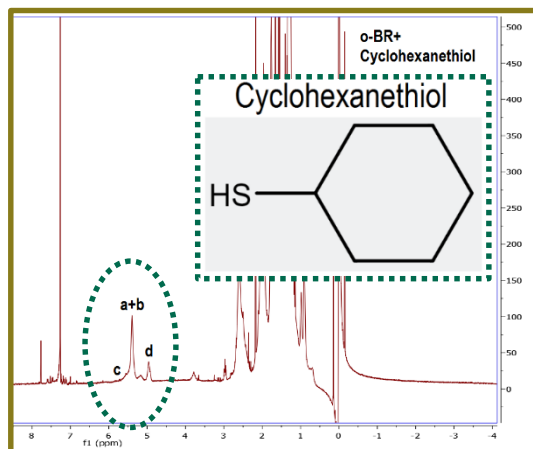
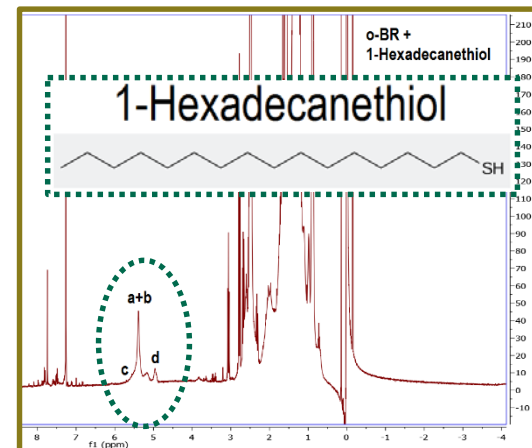
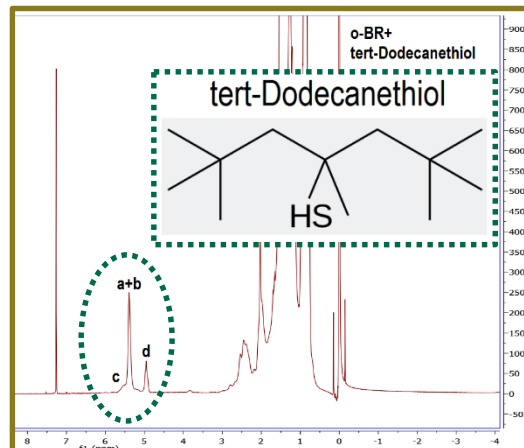
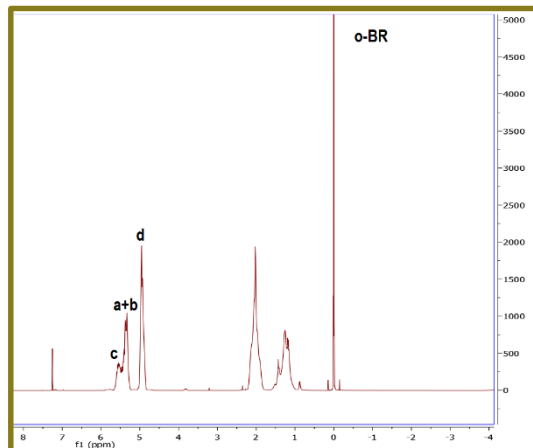
## OLIGOMER BRANCHING RESULTS ANALYSED BY LS-NMR





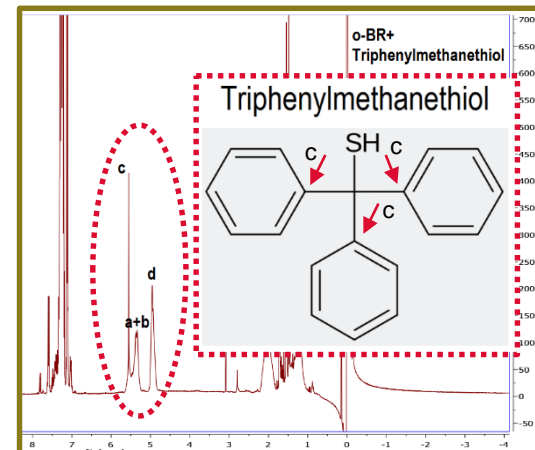
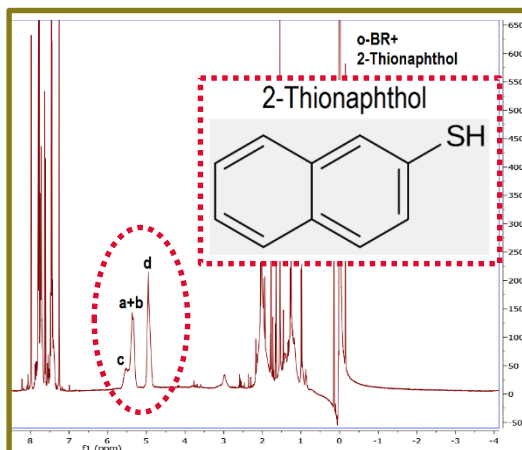
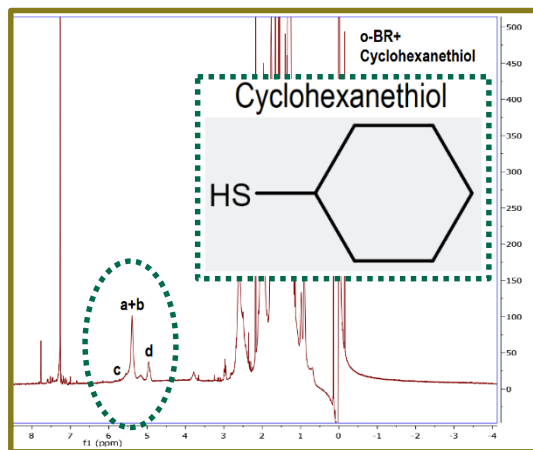
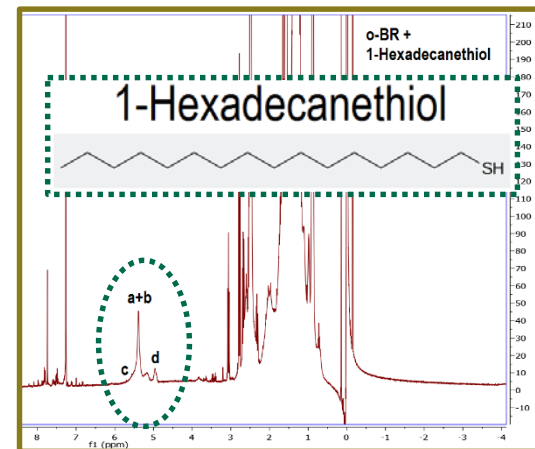
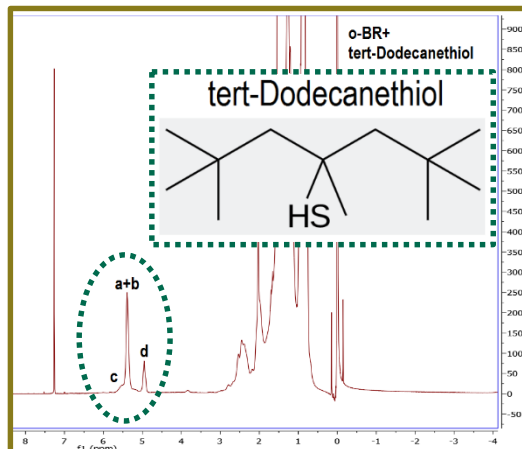
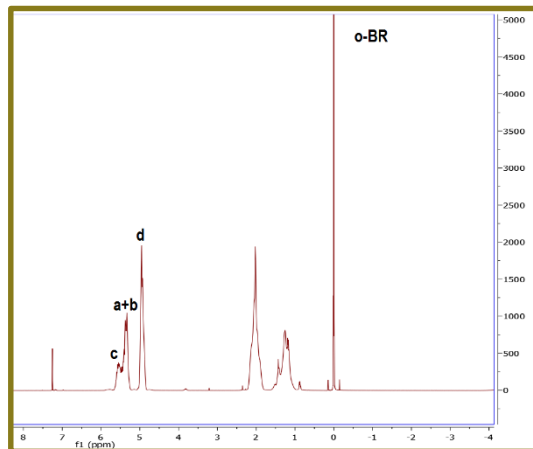
# BRANCHING OF OLIGOMER-BACKBONE

## OLIGOMER BRANCHING RESULTS ANALYSED BY LS-NMR



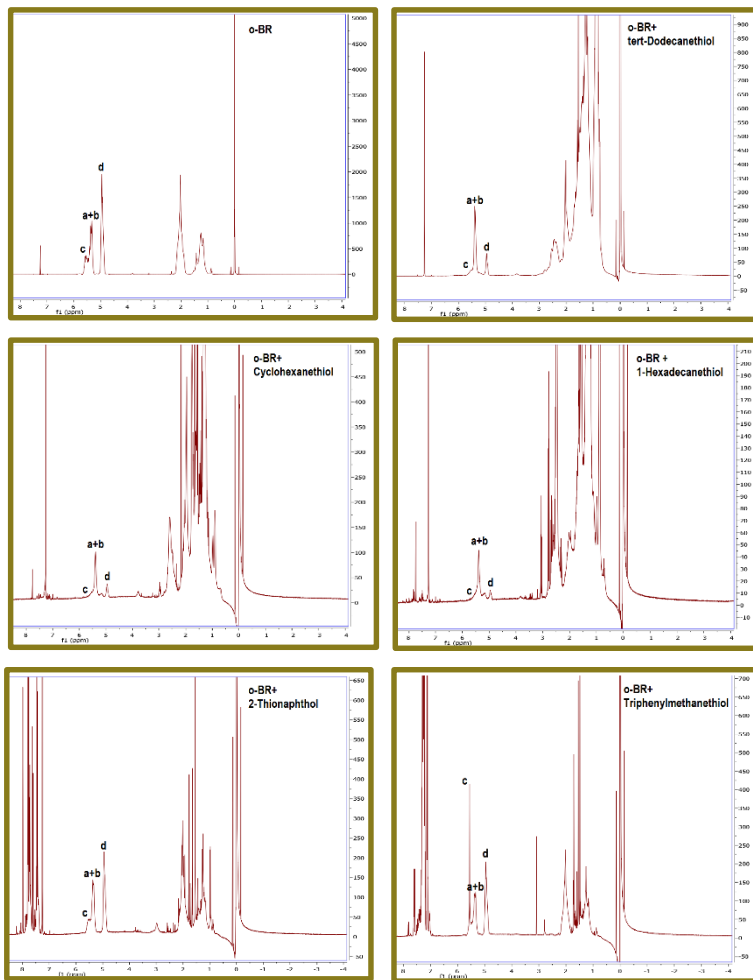
# BRANCHING OF OLIGOMER-BACKBONE

## OLIGOMER BRANCHING RESULTS ANALYSED BY LS-NMR



# BRANCHING OF OLIGOMER-BACKBONE

## OLIGOMER BRANCHING RESULTS ANALYSED BY LS-NMR



Presence of aromatic groups seems to influence **negatively** the effectiveness of the reaction.

Sample	Integration ratio (d/a+b)
o-BR	1.351
o-BR + tert-Dodecanethiol	0.287
o-BR + Cyclohexanethiol	0.427
o-BR + 1-Hexadecanethiol	0.319
o-BR + 2-Thionaphthol	1.099
o-BR + Triphenylmethanethiol	1.316

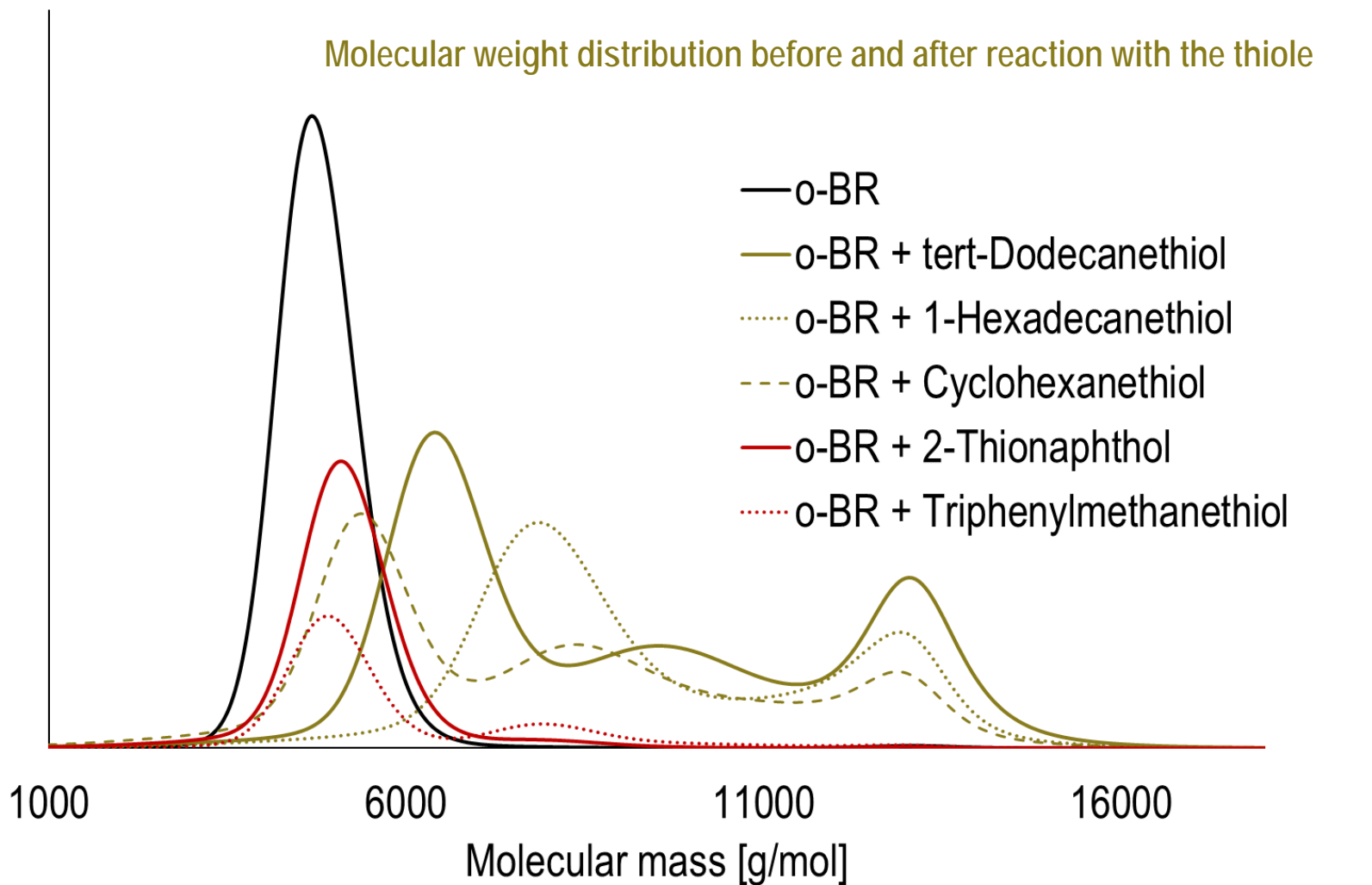
Possibly presence of **electron-donor alkyl group** is necessary for effective grafting to vinyl groups.

Side reaction – **recombination** of thiole radicals?

# BRANCHING OF OLIGOMER-BACKBONE

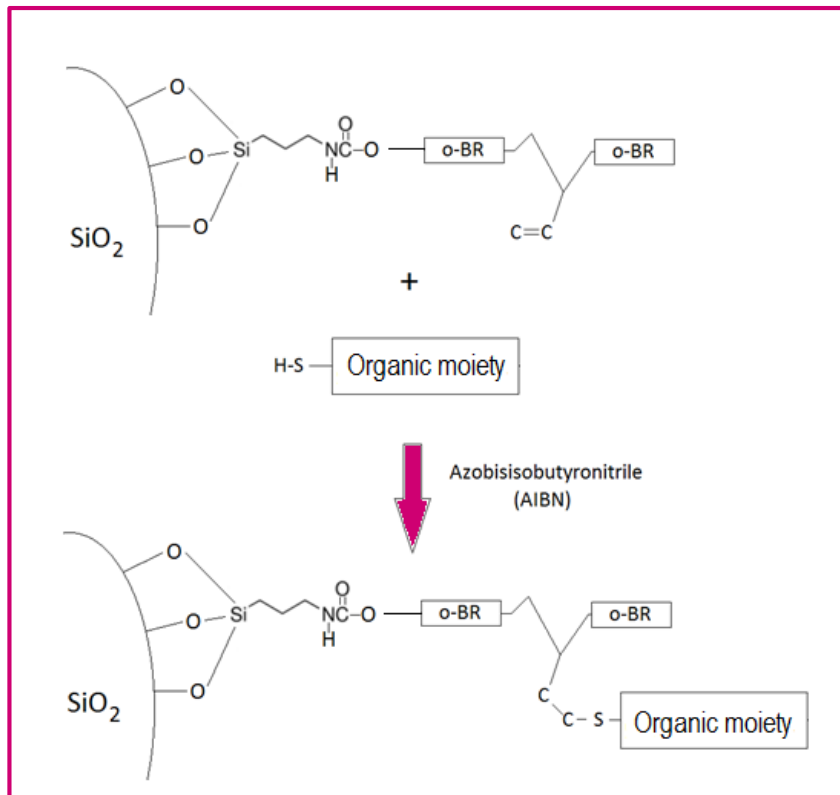
OLIGOMER BRANCHING RESULTS ANALYSED BY GPC

Molecular weight distribution before and after reaction with the thiole



# BRANCHING OF OLIGOMER-BACKBONE

## PROCEDURE CHARACTERISTICS



### Procedure parameters:

Duration – 1 hour

Temperature – 65-70 °C (constant growth)

Mechanical stirring – 150 rpm

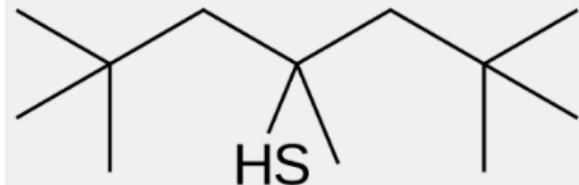
Atmosphere – nitrogen

Evaporation – 24 hours, 70 °C

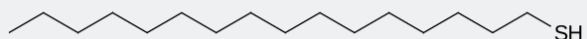
Degassing - 24 hours, 70 °C (vacuum)

### *Thiols used for oligomer-backbone branching*

#### tert-Dodecanethiol



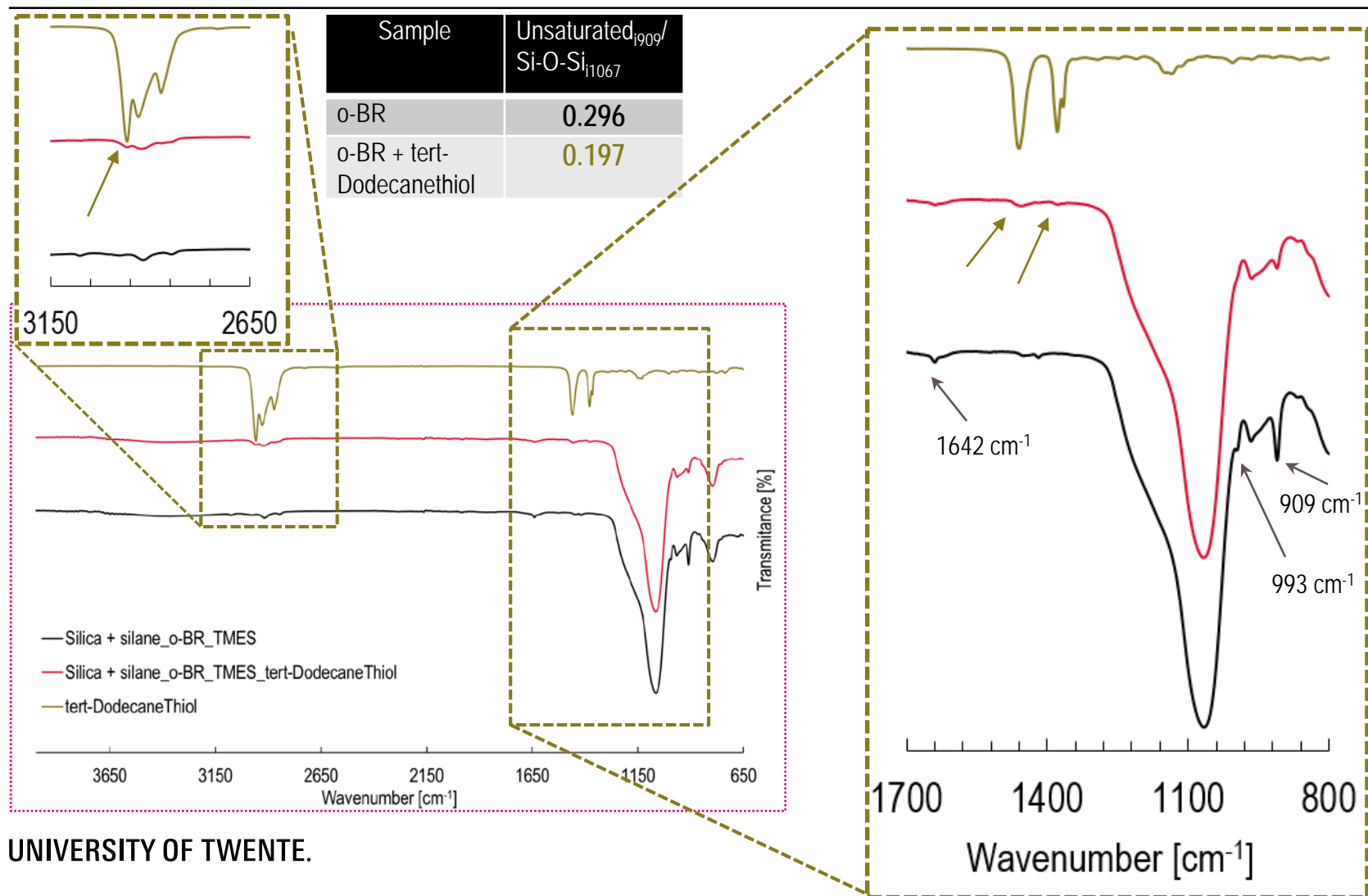
#### 1-Hexadecanethiol





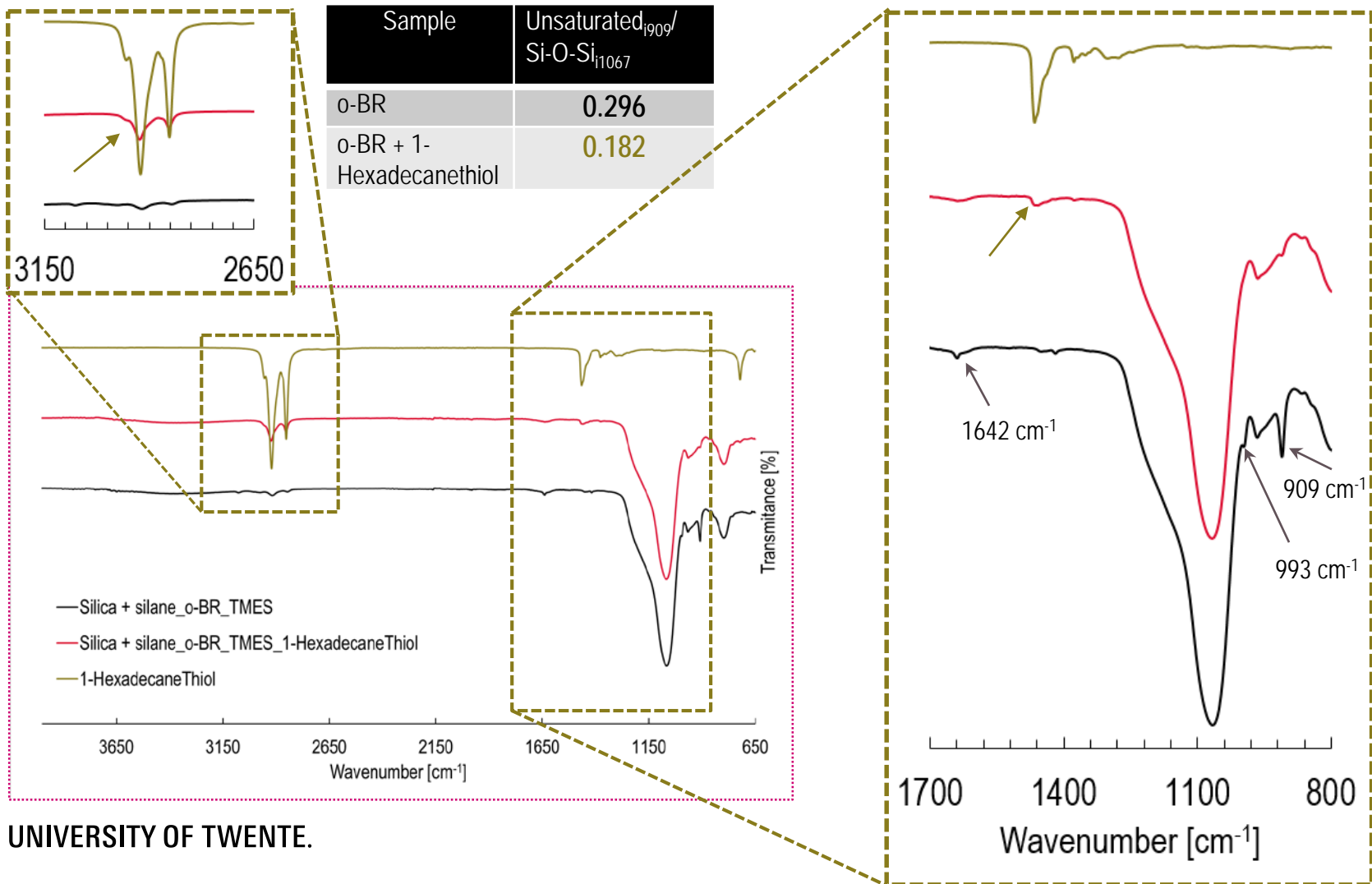
# BRANCHING OF OLIGOMER-BACKBONE

## GRAFTED-OLIGOMER BRANCHING ANALYSED BY FTIR



# BRANCHING OF OLIGOMER-BACKBONE

## GRAFTED-OLIGOMER BRANCHING ANALYSED BY FTIR

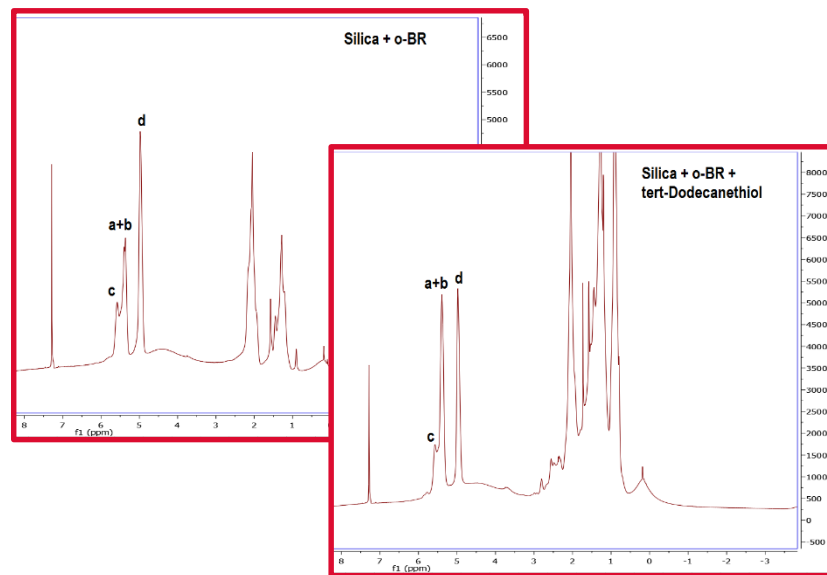
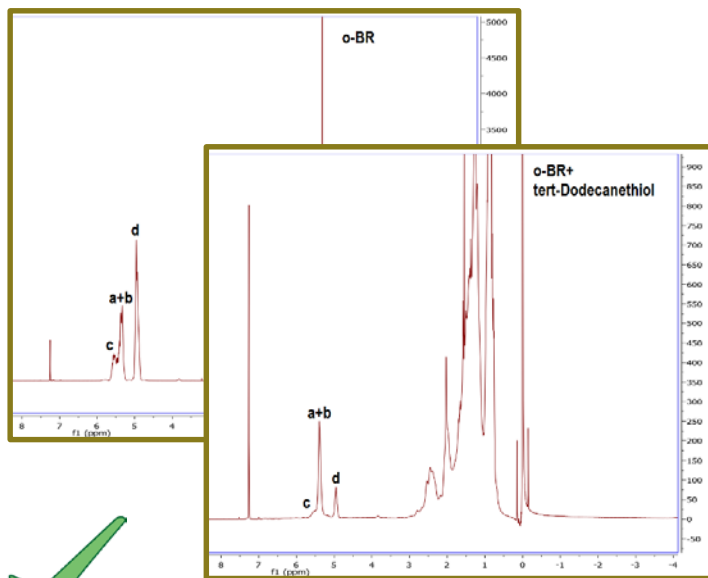


# BRANCHING OF OLIGOMER-BACKBONE

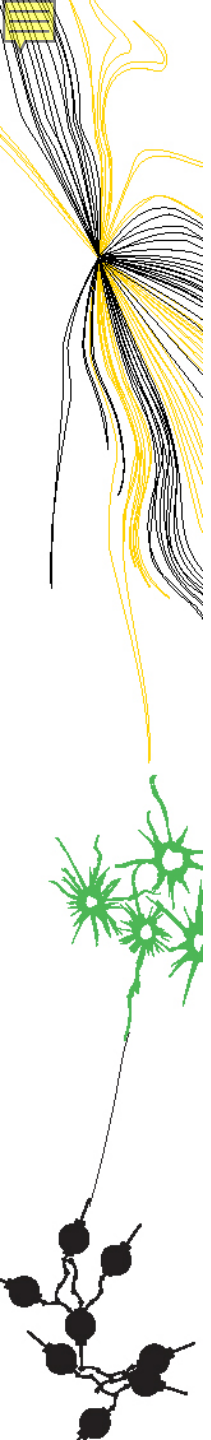
GRAFTED VS NON-GRAFTED OLIGOMER BRANCHING ANALYSED BY HR-MAS NMR

Sample	Integration ratio (d/a+b)
o-BR	1.351
o-BR + tert-Dodecanethiol	0.287
% reacted vinyl groups	78.8 %

Sample	Integration ratio (d/a+b)
Silica + o-BR	1.235
Silica + o-BR + tert-Dodecanethiol	0.901
% reacted vinyl groups	27.0 %



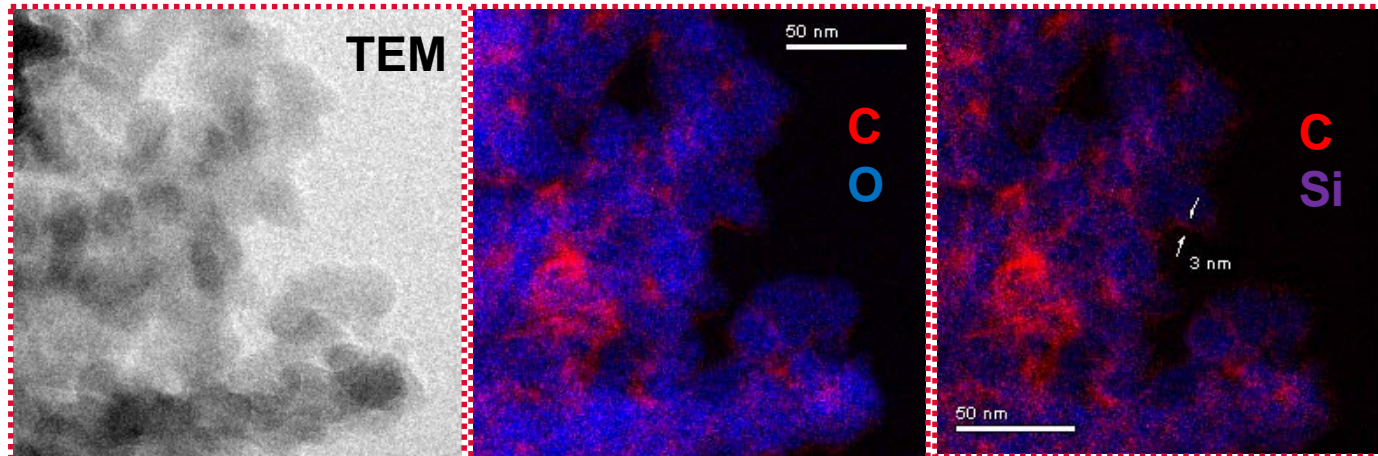
Negative influence of water?  
Immobilization of thiols on silica surface?



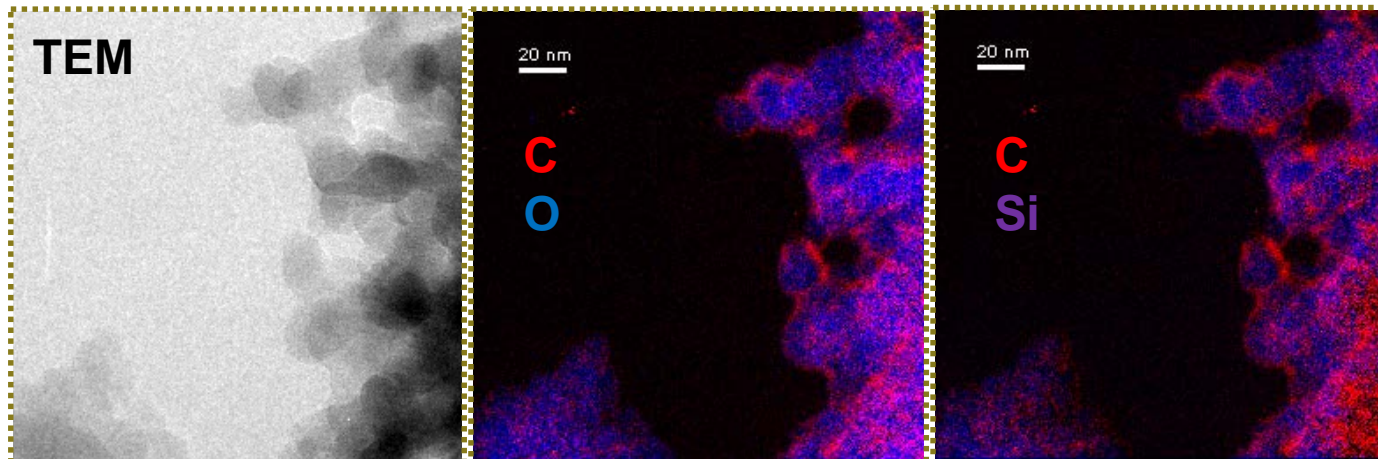
# BRANCHING OF OLIGOMER-BACKBONE

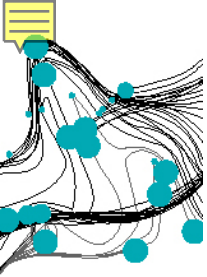
BRANCHED VS NON-BRANCHED OLIGOMER ON SILICA ANALYSED BY EFTEM

Non-branched o-BR on silica surface



O-BR branched with tert-dodecanethiol on silica surface





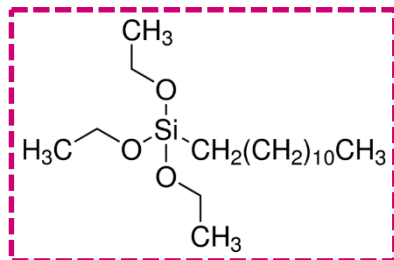
# GREEN MIXES PERFORMANCE

## PREPARATION OF SILICA FILLED SSBR GREEN MIXES

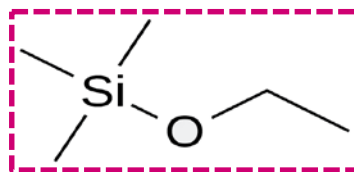
### SSBR 100 phr + 70 phr of:

- Silica + Trimethylethoxysilane (TMES)
- Silica + Dodecyltriethoxysilane (D-DTES)
- Silica + Isocy\_silane\_o-BR + tert-Dodecanethiol
- Silica + Isocy\_silane\_o-BR + 1-Hexadecanethiol

D-DTES

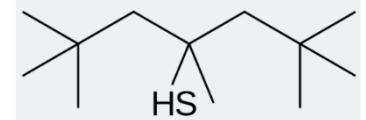


TMES

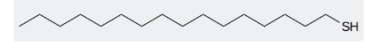


### Thioles

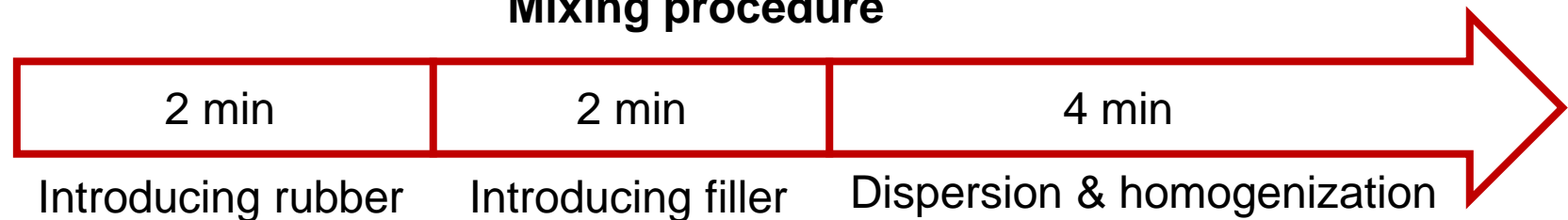
tert-Dodecanethiol



1-Hexadecanethiol



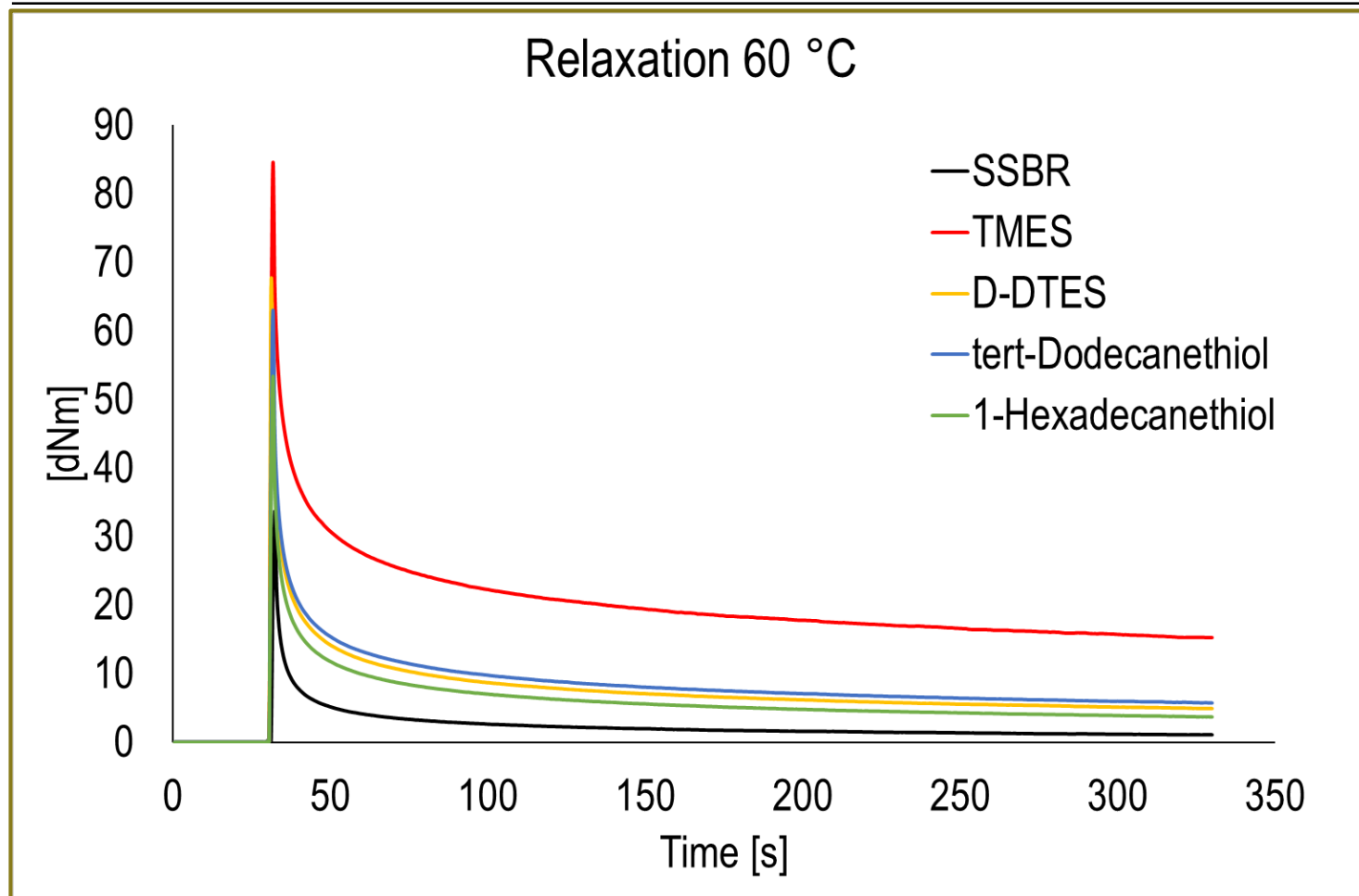
### Mixing procedure





# GREEN MIXES PERFORMANCE

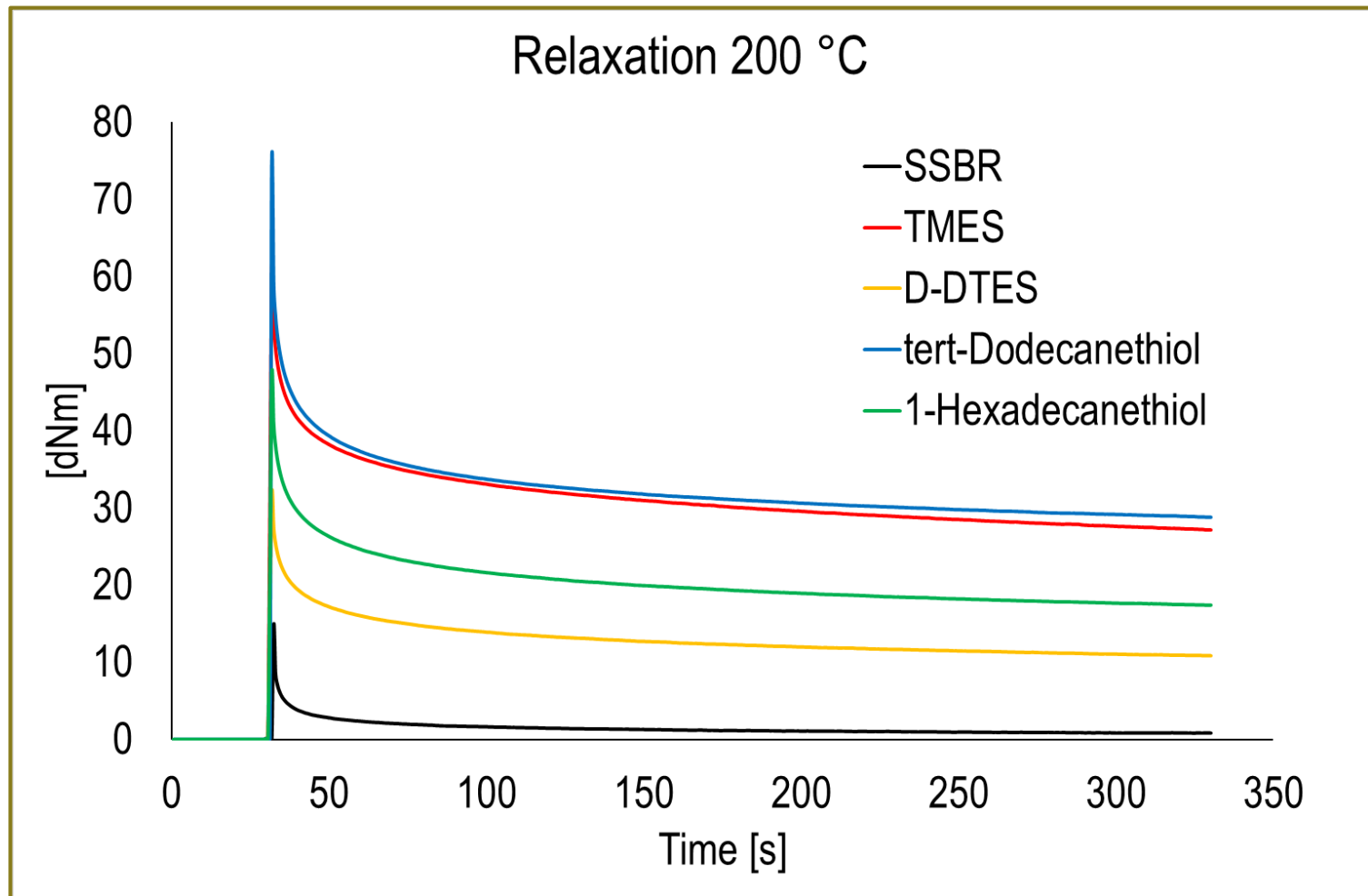
PROPERTIES OF THE MIXES

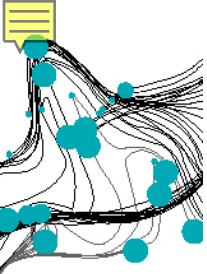




# GREEN MIXES PERFORMANCE

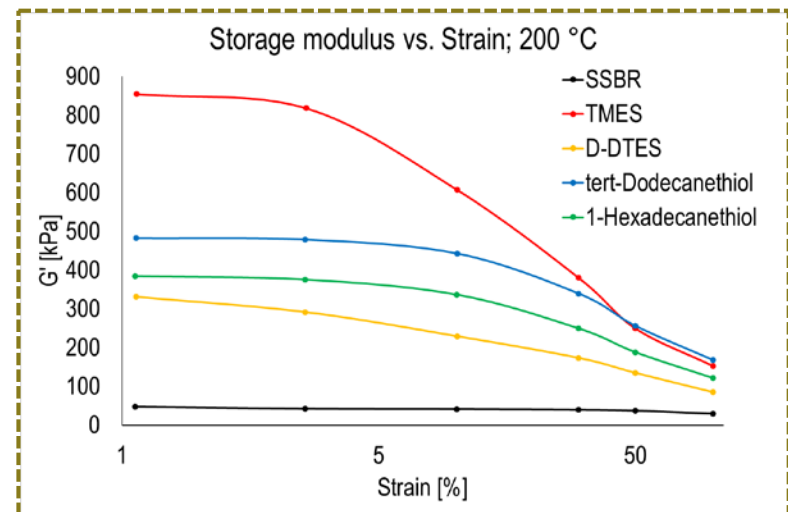
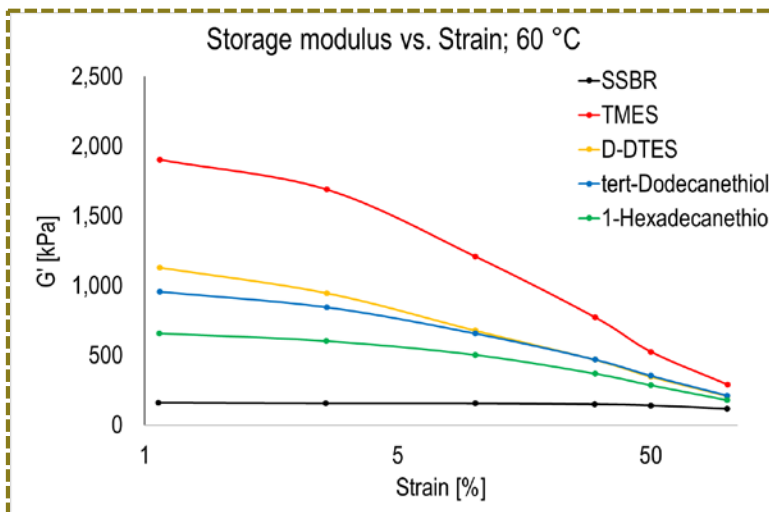
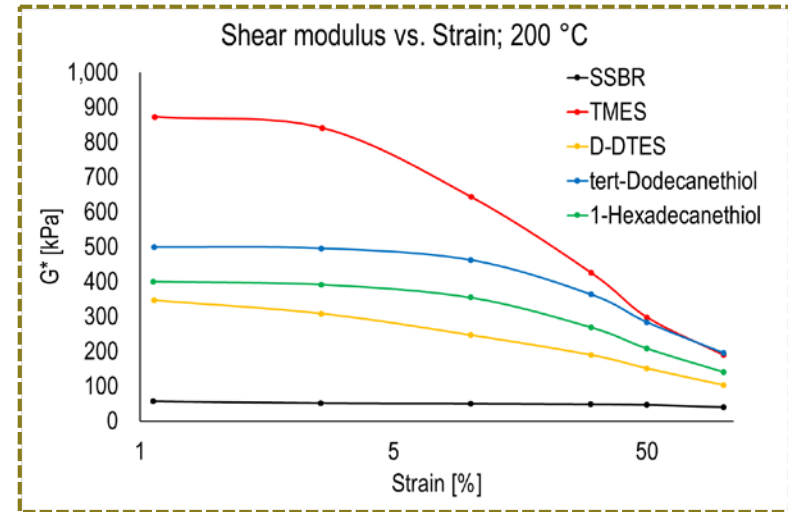
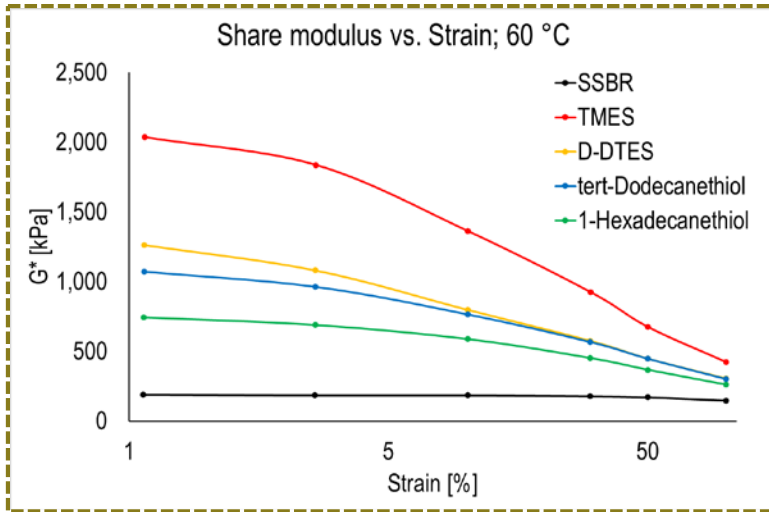
PROPERTIES OF THE MIXES



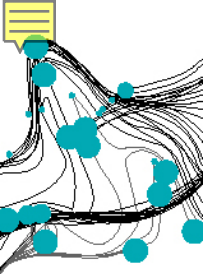


# GREEN MIXES PERFORMANCE

## PROPERTIES OF THE MIXES

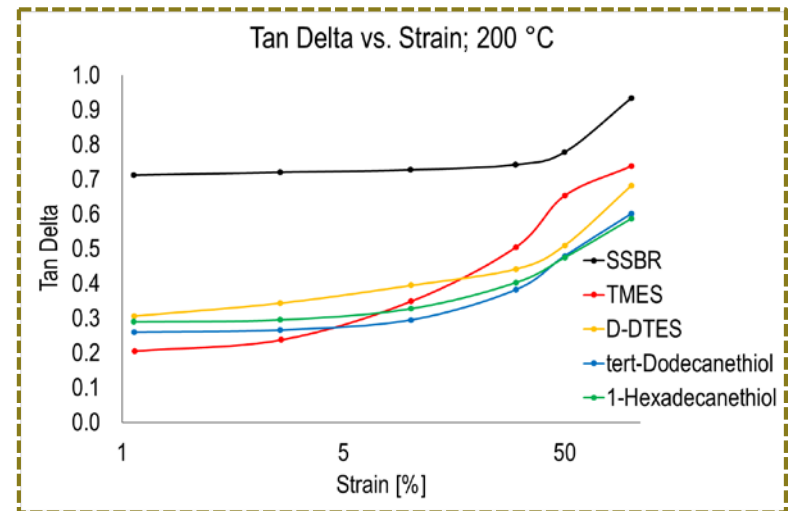
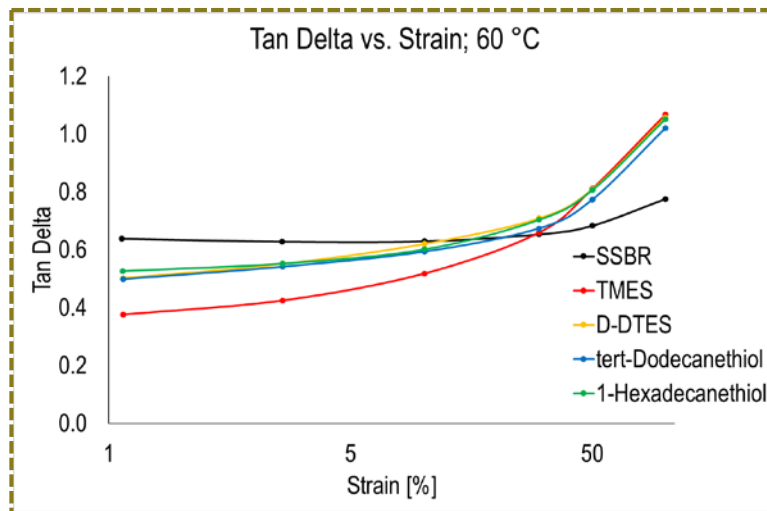
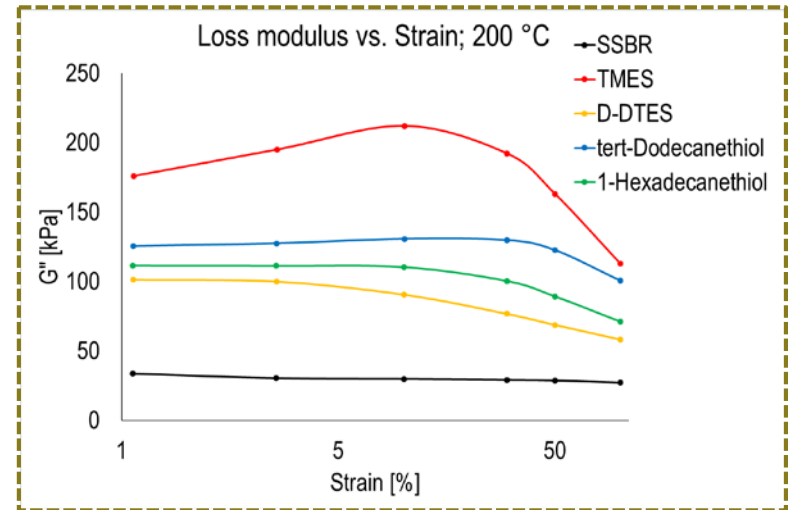
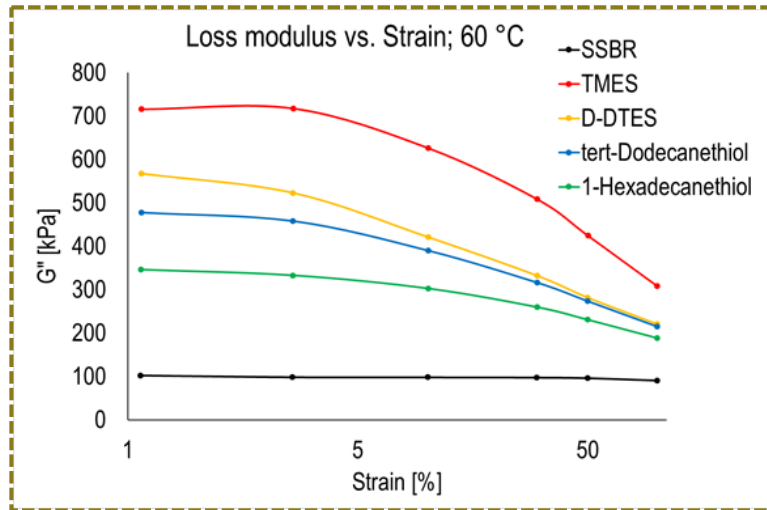






# GREEN MIXES PERFORMANCE

## PROPERTIES OF THE MIXES





# SUMMARY

## NEXT STEPS

---

### **O-BR modified with small-molecular thioles:**

- TGA analysis of silica modified with the various compounds
- Enhancing efficiency and analysis of thioles reaction with vinyl groups of the oligomer backbone grafted and not-grafted on silica surface

### **Green mixes filled with the modified silica:**

- Preparation of rubber samples filled with silica covered with o-BR and Cyclohexanethiol modified o-BR
- DMA analysis of the samples
- SEM and DisperGrader analysis of the samples



# SUMMARY

## CONCLUSIONS

---

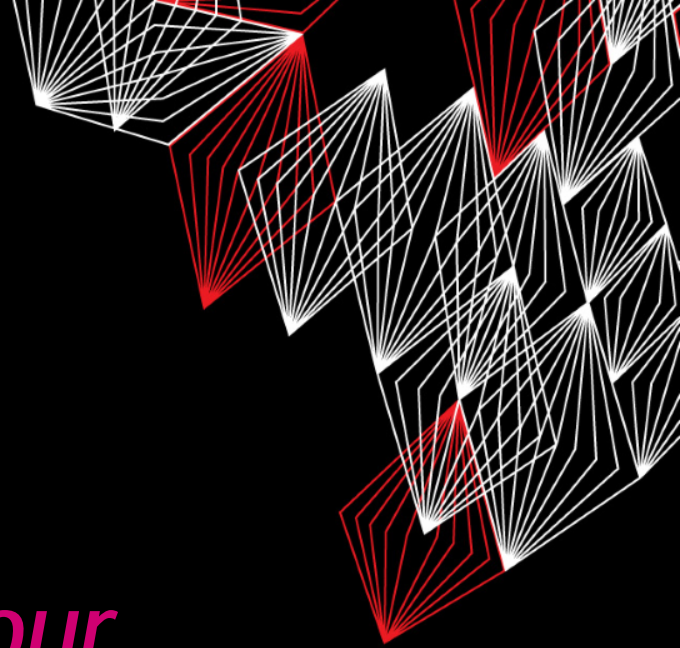
### Conclusions:

- Reaction between telechelic monohydroxy-butadiene oligomer (o-BR) and isocyanate silane allows grafting of relatively large organic chains on silica surface with high efficiency.
- Utilization of polybutadiene backbone containing vinyl groups enables effective branching of the macromolecule with various thioles.
- Developed procedure provides a simple and effective method of long branched-molecules grafting on silica surface.
- Addition of modified silica to SSBR rubber results in interesting dynamic properties, especially at elevated temperature when macromolecular mobility is high.

UNIVERSITY OF TWENTE.

3 July 2018,  
Nuremberg, Germany

*Thank you for your  
kind attention!*



**DKT 2018**  
Deutsche Kautschuk-Tagung

