Data Article

Data on laser induced preferential crystal (re) orientation by picosecond laser ablation of zinc in air

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Abstract

Laser ablation of zinc is performed with a 6.7 ps pulsed laser source to investigate the ablation mechanism and resulting morphology of the irradiated surface. The data shows the changes in crater morphology, as well as chemical composition, for different number of pulses and laser fluence levels. We observed Laser Induced Preferential Crystal Orientation (LIPCO), as a result of ultra-short pulsed laser processing of Zn at a wavelength of 515 nm. Crystallographic data for other laser wavelengths, namely 343 and 1030 nm, as well as for Zn coated steel are also provided in support of this observation. Data presented in this article are related to the research article "Investigation of the ultrashort pulsed laser processing of zinc at 515 nm: morphology, crystallography and ablation threshold" [1].

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1. Data

1.1. Crater morphology

The degree of micro- and nano-particle deposition around a given crater is found to qualitatively increase with increasing ablated depth. Fig. 1 shows representative craters at all the experimental conditions. The yellow line separates the craters around which remarkable particle redeposition occurs, while the blue line demarcates the craters that show a so-called “drilling effect”. In the context of this work, when the ratio between the depth and the diameter of the crater \( h_{\text{crater}}/D_{\text{crater}} < 0.2 \) and the ratio of crater diameter to beam diameter \( D_{\text{crater}}/2\omega_0 < 1.5 \), the crater morphology is said to demonstrate this drilling effect. Also in this case, the maximum depth of the craters is greater than 8.5 mm. This is shown in Fig. 2 with the dashed horizontal line. The solid horizontal line at 4 mm represents the depth above which micrometric particle redeposition becomes prominent.

1.2. Chemical composition

Both EDS and XPS measurements have been performed on laser-induced craters. However, no significant difference in the chemical composition of Zn, Al, C and O was observed. For XPS measurements, the average carbon concentration for all laser processing conditions is approximately 50%, with C1s binding energy of 284.8 eV, which is attributed to adventitious carbon [2]. Fig. 3 shows the elementwise concentration of Zn, Al and O by offsetting the C concentration. In this figure, the graph on the left is for \( F_0 = 2.4 \text{ J/cm}^2 \) for number of pulses ranging from \( N = 1 \) to 50, while the graph on the right shows the atomic concentration for single pulse (\( N = 1 \)) at fluence levels up to 35 J/cm².

1.3. Crystallography

For Zn, laser induced preferential crystal orientation (LIPCO) is not limited to the laser wavelength of 515 nm, for picosecond laser pulses. Similar reorientation of the modified area was also observed for
laser wavelengths of 1030 and 343 nm as shown in Fig. 4. This suggests that this physical phenomenon does not strongly depend on laser wavelength. However, (so long as the irradiated grain is not already in the preferred orientation), for 1030 nm, partial (preferential) re-orientation is found, whereas for wavelengths of 515 and 343 nm full reorientation is observed, which can be seen in Fig. 4. It appears that the flatter the crater bottom gets by melt expulsion, the higher the degree of preferred orientation is observed. More details of LIPCO and proposed physical driving mechanism are discussed in Ref. [1].

Crystal (re)orientation, i.e. re-orientating from any crystallographic plane to (wards) ⟨0001⟩, was also observed for Zn coated steel (i.e. galvanized steel) processed at a wavelength of 515 nm as shown

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**Fig. 1.** SEM micrographs (top view) of zinc surface irradiated at different laser peak fluence $F_0$ [J/cm$^2$] levels (rows) and at different number of laser pulses $N$ (columns). Yellow and blue lines mark the regions with prominent micrometric droplets around the crater and drilling effect respectively. All images are in the same scale, where the white scale bar indicates 10 μm.
in Fig. 5(b). However, we did not observe this phenomenon (LIPCO) for craters produced with a wavelength of 1030 nm (see Fig. 5(a)). At this wavelength, it is difficult to observe LIPCO, because of the surface roughness (non-indexed points [3]) and crater depth (shadowing effect [4]). Moreover, the surface chemical composition of galvanized steel is different than pure Zn due to the presence of an Al rich oxide film, which results from precipitation during the hot dip galvanizing process [5]. Therefore,
slight differences in surface chemistry or stress in the Zn crystals might also have an influence on the degree of crystal (re)orientation. In any case, more research is required to study whether LIPCO also occurs (to which extend, or not) at other wavelength, other laser-conditions and for other Zn-based alloys.

2. Experimental design, materials and methods

Laser ablation experiments were performed in a similar manner as mentioned in Ref. [1]. Along with the second harmonic wavelength, the fundamental (1030 nm) and the third harmonic (343 nm) wavelengths were also used in this work. The focal spot (1/e2) radius were measured using a MicroSpot Monitor (Primes GmbH, Germany) and found to equal 14.4 ± 1.6 μm, 11.9 ± 1.6 μm and 7.4 ± 0.5 μm with an ellipticity 0.93, 0.78 and 0.86 for a wavelength of 1030, 515 and 343 nm respectively.

Pure zinc samples were similar to those mentioned in Ref. [1]. The coated samples (galvanized steel) with a surface roughness (Sₚ) of 0.3 μm is commercially produced according to European standard EN10346:2015 and has a nominal Zn layer thickness of 10 μm. While pure zinc samples were cleaned with ethanol in an ultrasonic bath, the coated samples were cleaned (swabbing) using Ammonia (<5%) solution prior to and after the ablation experiments.

The dimensional, morphological and crystallographic measurements using confocal laser scanning microscope (CLSM), scanning electron microscope (SEM) and electron backscatter diffraction (EBSD) respectively, were performed as mentioned in Ref. [1]. X-ray Photoelectron Spectroscopy (XPS), (Quantera SXM of Physical Electronics, USA) was used to analyze the chemical composition of the pure Zn samples at similar conditions mentioned in Ref. [6].
Acknowledgements

The authors would like to acknowledge the financial support of Tata Steel Nederland Technology BV. We would also like to thank G. Kip of MESA+, University of Twente for his help with the XPS measurements.

Transparency document

Transparency document associated with this article can be found in the online version at https://doi.org/10.1016/j.dib.2019.103922.

References
