

Atomic Layer Deposition of W- based layers on SiO₂

S. Bystrova¹, J. Holleman¹, R. A. M. Wolters², A. A. I. Aarnink¹

University of Twente¹, Faculty of Fac. of Electrical Engineering, Mathematics & Computer Science (EEMCS), Semiconductor Devices Group (ICE), P.O.Box 217, 7500 AE Enschede, The Netherlands

Philips Research², The Netherlands

Phone: +31 (0)53-489 27 29 Fax: +31 (0)53-489 10 34

E-mail:S.Bystrova@utwente.nl

Abstract—W<Si> and W_{1-x}N_x, where x= 15- 22 at%, thin films were grown using the ALD (Atomic Layer Deposition) principle. Growth rate of W<Si> films is about 4- 5 monolayers/ cycle at 300- 350 °C. Growth rate of W_{1-x}N_x is 0.5 monolayer/cycle at 325- 350 °C. Standard Deviation (STDV) of thickness is about 2% for 20nm layers. Specific resistivity is 180 mW cm for W<Si> and as low as 220-340 mW cm for W_{1-x}N_x 20nm films. 4 point probe sheet resistivity test is applied to Cu/ Barrier/ SiO₂ stacks with ramping temperature. No changes of normalized resistance reflecting Cu-Barrier interaction was measured after 500 °C annealing cycle.

Keywords— Atomic Layer Deposition; tungsten; copper metallization; diffusion barrier; tungsten nitride.

I. INTRODUCTION

Diffusion barriers are required in silicon IC's with Cu interconnects to protect Si and dielectrics against Cu-diffusion and Cu against diffusion of Si [1]. There are some major geometrical and physical requirements for barriers. First of all, the thickness of the diffusion barrier should be 10 nm in 2004 and 5 nm in 2010, according to International Technology Roadmap. The conventional ionised Physical Vapour Deposition (iPVD) is unable to deposit such a thin layer conformally over features with high aspect ratio. The Atomic Layer Deposition (ALD) method, based on the chemisorption of gas- precursors, allows to decrease the growth rate down to =1 monolayer/ cycle and to deposit films with 100% step coverage. References in the literature about ALD of W- based films on dielectrics are scarce, although W_xN_yC_{1-x-y} films grown by ALD have been reported to be promising barriers referring to low resistance and

morphology [2]. Clearly, more research should be done in this field for understanding of the growth mechanism of the layers.

II. EXPERIMENTAL PROCEDURE

A. ALD equipment

Deposition experiments are carried out in ALD-system. The Si-wafer is placed on a Mo- susceptor in a load- lock. When the load- lock is pumped down to 1*10⁻⁵ mbar, the Mo- susceptor is transferred into a hot reactor and placed on the bottom heater. The bottom heater goes up forming a closed hot- wall reactor. The reactor volume is 24 ml. The base pressure in the reactor is 2.7* 10⁻⁷ mbar.

B. Wafer preparation and deposition experiments

Prior the deposition, wafers with 700nm thermal oxide grown in water vapor were cleaned with a standard cleaning in fuming nitric acid and boiling 69% HNO₃, rinsed with demi- water. Wafers were dipped into 0.3 % HF for 5min., respectively, and rinsed in the demi- water to form -OH groups on the surface of SiO₂. These hydroxyl groups serve as reactive sites to initiate the film growth. When the wafer was loaded into the reactor, a Si- nucleation layer was grown by decomposition of Si₂H₆.

W- films were deposited with a sequence of Si₂H₆/WF₆ pulses, N₂ was used as a purging gas. A sequence of WF₆/ NH₃/ C₂H₄ /SiH₄/ NH₃ was applied to deposit W_xN_{1-x} films. Experiments were carried out

at 1 mbar. Variable parameters were temperature, pulse time, number of cycles. The temperature range was 250- 350 °C. N₂ purging time was 2sec. The cycle time was 7.5- 11 sec for W and 14- 22 sec for W_xN growth. The amount of reactive gas was controlled by mass- flow controller for WF₆, Si₂H₆, NH₃, SiH₄ and C₂H₄ at 5 sccm with a total flow in the reactor of 45- 80 sccm.

The composition of the films grown was measured with XPS (X- Ray Photoelectron Spectroscopy) using standard sensitivity factors. Surface roughness was measured with AFM (Atomic Force Microscopy). The thickness was calculated from the increase of the weight using the standard density for W of 19.3 and for W₂N of 17.2 g/cm³.

C. Details of the resistivity test

A 4- point probe sheet resistivity test was performed in vacuum at a base pressure of 4*10⁻⁶ mbar at elevated temperatures. The measurements are done with a ramping up and down step of 0.01degree/ sec. The test includes the heating up step to 500 °C for 30 minutes step at constant temperature and cooling down to room temperature. Stacks of Cu(100nm)/ Barrier/ SiO₂ and Cu(200nm)/ Barrier/ SiO₂ grown on Si- substrate are used for the test. Cu was deposited on wafers by rf-sputtering.

III. RESULTS AND DISCUSSIONS

A. W growth in the Si₂H₆/ WF₆ system

W deposition was studied at the temperature range 250- 350 °C. No temperature window for the deposition of W in ALD mode was found. Results of the experiments are shown in figure 1.

The growth of W films was about 3 to 5.7 monolayers/ cycle. Obviously, Si₂H₆ decomposes on the fresh W- surface, which is a well- known catalyst, with the formation of more than 1 monolayer of Si/cycle. All Si formed is consumed by the following WF₆ pulse. The adsorption of Si₂H₆ and high reactivity of WF₆ with Si are determining factors of a high growth rate.

The role of the Si- layer in the growth process was investigated at 325 °C. The thinner the nucleation layer is, the slower the growth is on initial stages and the incubation time is longer, see figure 2. When the surface is converted into W, the growth rate for

processes with a varying nucleation step is almost the same. The growth rate calculated from the slope of the growth curves is 4.5 monolayer/cycle for the Si₂H₆/ WF₆ ratio equal to 4/3 and 2/1.5. The growth rate is 3.6 monolayer/ cycle for the Si₂H₆/ WF₆ ratio equal to 1.4/2. This means that the nucleation layer helps to decrease the incubation time, but will not influence the final growth rate. The smaller incubation time of the growth the smoother the growing film is. AFM test showed roughness as high as 2.7 nm for 10 nm layer without nucleation and 0.52 nm for the nucleation at 2.3*10⁷ L.

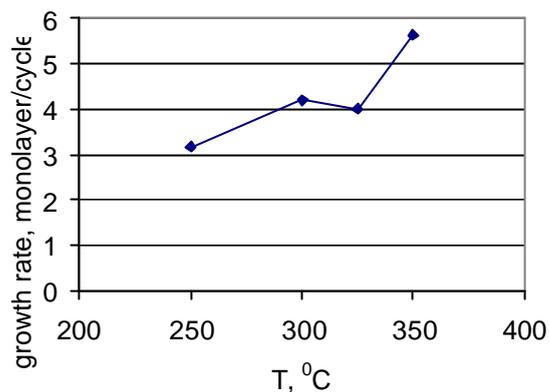


Figure 1. Temperature dependence of the W<Si> growth with a ratio of Si₂H₆ / WF₆ = 4/3 after 20 cycles, Si-nucleation layer was grown at exposure to 4.5*10⁹ L (Langmuir, 1L= 10⁻⁶ Torr*sec) of Si₂H₆. Growth rate was calculated from the total weight increase.

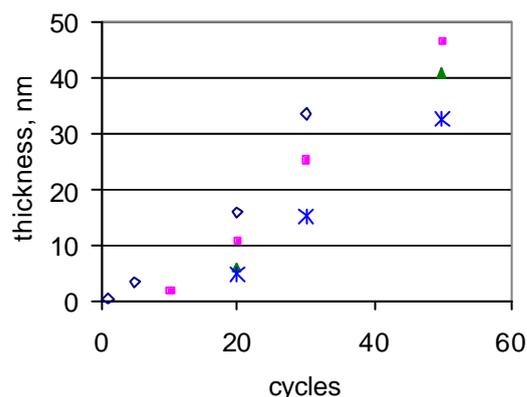


Figure 2. Thickness of W versus number of cycles at 325 C. In the legend:

- is for Si₂H₆/WF₆= 4/3, nucleation at 4.5*10⁹L of Si₂H₆;
- is for Si₂H₆ / WF₆= 4/3 and nucleation at 2.3*10⁷ L;
- ▲ is for Si₂H₆ / WF₆= 2/1.5 and nucleation at 2.7*10⁵ L;
- * is for Si₂H₆ / WF₆= 1.4/2 and nucleation at 2.7*10⁵ L.

The resistivity of the grown films was in the range 160- 200 $\mu\Omega\text{cm}$ for 25 nm films. The resistivity does not differ much with varying $\text{WF}_6/\text{Si}_2\text{H}_6$ ratios, however the uniformity of the films does change. The best results based on STDV of resistivity were obtained for the $\text{WF}_6/\text{Si}_2\text{H}_6$ ratio equal to 3/ 4 with a STDV of about 2 %, see figure 3.

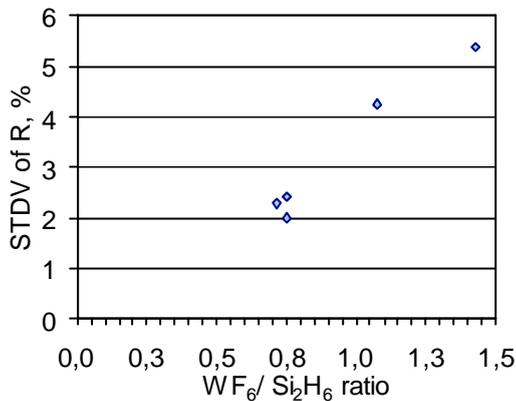


Figure 3. STDV of sheet resistance versus the ratio of WF_6 to Si_2H_6 for films with a thickness about 15 nm. The nucleation step is $4.5 \cdot 10^9$ L.

The deposited layers contain about 12 at% of Si according to XPS.

B. Growth of W_xN_x films in the $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4$ system

W_{1-x}N_x is reported to have a high resistivity of about 4500 $\mu\Omega\text{cm}$ [3]. $\text{W}_x\text{C}_y\text{N}_{1-x-y}$ layers possess a resistivity of 300- 400 $\mu\Omega\text{cm}$ with STDV of 3.5 % for 25nm films[2]. The C_2H_4 was chosen as a cheap carbon precursor and as an alternative to triethylborane (TEB). No continuous films were deposited with pulse sequences of $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4$, $\text{WF}_6/\text{C}_2\text{H}_4/\text{NH}_3$. An extra step with a SiH_4 pulse (a sequence of $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4$) resulted in roughness of 1.7 nm for a 20 nm film, due to direct interaction of Si with WF_6 . That is not acceptable for a barrier application. A pulse sequence of $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4/\text{NH}_3$ was applied for the growth of W_{1-x}N_x layers to prevent a direct interaction of Si with tungsten fluoride. No carbon and Si are incorporated however in the growing films. C_2H_4 and SiH_4 play a sacrificial role and all products formed with carbon and Si are removed. The effect of the

C_2H_4 exposure on the films growth is not clear yet, Table 1, N 5-6.

Table 1. Summary table of the final growth rate and incubation time at different conditions

N	$\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4/\text{NH}_3$	$T, ^\circ\text{C}$	incubation time, cycles	growth rate, monolayer/cycle
1	1/ 2/ 1/ 2/ 2	325	0,7	0,47
2	2/ 2/ 1/ 2/ 2	325	5,3	0,55
3	1/ 2/ 1/ 2/ 2	*4,5e9 L	no	0,47
4	1/ 2/ 3/ 2/ 2	*4,5e9 L	no	0,57
5	1/ 2/ 3/ 2/ 2	350	no	0,51
6	1/ 2/ 1/ 2/ 2	350	no	0,50
7	3/ 2/ 3/ 2/ 2	350	3,9	0,68

*) Process after a nucleation step with a large exposure to Si_2H_6 equal to $4.5 \cdot 10^9$ L

The temperature window for the W_{1-x}N_x growth was limited to 325- 350 $^\circ\text{C}$. It is not possible to go to lower temperature because of the formation of a WF_xNH_3 adduct [4]. On the other hand, 350 $^\circ\text{C}$ is the maximum possible temperature in the back-end for Cu/SiLK combination. The growth rate was about 0.5- 0.6 monolayer/ cycle for a varying pulse time of C_2H_4 and WF_6 , figure 4. The growth rate was calculated from the slope of thickness- cycles dependence, see figure 5.

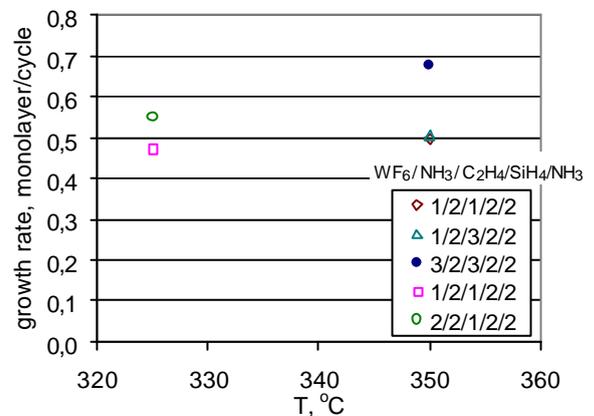


Figure 4. Growth rate of W_{1-x}N_x versus temperature with varying pulse sequence of $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4/\text{NH}_3$. Time of reactive pulses is in the legend.

The thickness of initial Si film was the same for both temperatures (1.2 nm for 325 $^\circ\text{C}$ and 1.5nm for 350 $^\circ\text{C}$) to equalize the starting conditions. Some experiments were performed after large initial exposure to Si_2H_6 $4.5 \cdot 10^9$ L, figure 5. The thickness of the nucleation

layer influences how fast the growing film becomes continuous, but not the final growth rate. Incubation time of the growth, calculated from the growth rate, see Table 1, was very short or, in some cases, the growth started almost immediately. The short incubation time resulted in a roughness as low as 0.43- 0.76 nm for 16 nm $W_{1-x}N_x$ films. The films possess extremely low resistivity up to 340 $\mu\Omega\text{cm}$ (STDV 2- 5%) for 10 nm films, compared to reported values [3].

The deposited films are $W_{1-x}N_x$ compounds with $x=15-22\text{at}\%$. Nitrogen suffers from the preferential sputtering during XPS profile measurements [5], therefore the initial part of the obtained profile is more useful to estimate a real nitrogen concentration. No Si peak was measured at the Barrier/ SiO_2 interface.

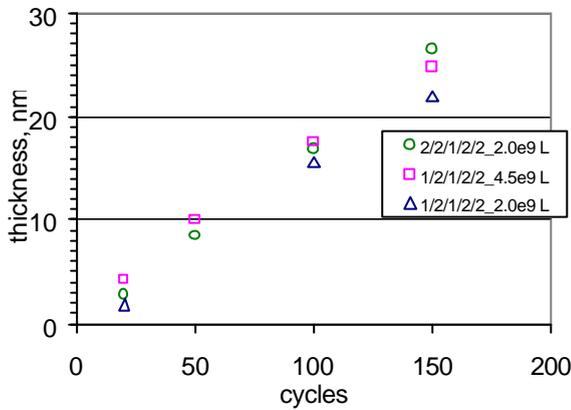


Figure 5. Thickness of the $W_{1-x}N_x$ films versus cycles at 325 °C. Ratio of gas reactants with a sequence of $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4/\text{NH}_3$ is in the legend.

C. Resistivity test

Stacks of Barriers on SiO_2/Si with 200 nm Cu layer show no change in normalized resistance after heating up to 500 °C and cooling down to room temperature. Resistivity of Cu was 2.1 $\mu\Omega\text{cm}$. In contrary, samples with as deposited 100 nm copper film on SiO_2 are characterized by a resistivity of 2.53 $\mu\Omega\text{cm}$. Such an increase of specific resistivity is caused by the amount of defects in the Cu layer. The resistance of each sample is decreased after the test, Table 2 and figure 6. This lowering of resistivity corresponds to annealing of the defects in the thin Cu- layer. The resistivity approached 2.1 $\mu\Omega\text{cm}$ and the TCR became 0.0028 deg^{-1} for each sample up to 100 °C. Thus no reaction

between Cu and Barriers occurred and the Cu/Barrier combination survived the 500 °C annealing test.

Table 2. Samples for the 4- point probe resistivity test with 100 nm Cu layer

N	barrier thick, nm	composition by XPS	Rbefore, $\text{m}\Omega/\text{sq}$	Rafter, $\text{m}\Omega/\text{sq}$	Rafter/Rbefore
1		Cu/ SiO_2	253	206	0,81
2	25,8	$W_{83}N_{17}$	450	225	0,50
3	20,5	$W_{83,5}N_{16,5}$	335	214	0,64
4	17	$W_{78}N_{22}$	365	235	0,64
5	26,3	$W_{84,5}N_{15,5}$	372	220	0,59
6	14,5	$W_{83,1}Si_{11,2}O_{5,8}$	310	206	0,66

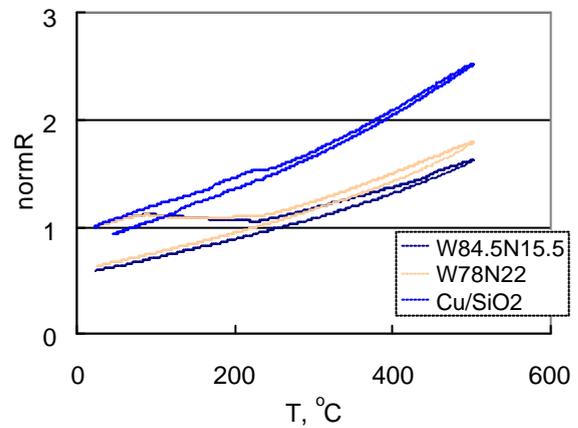


Figure 6. Normalized resistance change versus temperature during the measurements.

IV. CONCLUSIONS

No ALD conditions for W growth are found in the $\text{WF}_6/\text{Si}_2\text{H}_6$ system. Deposition rate is about 3-5 monolayers/ cycle at 250- 350 °C.

The Content of Si in these W films was about 11-12at%. Resistivity was up to 200 $\mu\Omega\text{cm}$ with STDV about 2% for 25 nm film

The growth rate in the $\text{WF}_6/\text{NH}_3/\text{C}_2\text{H}_4/\text{SiH}_4$ system was about 0.5- 0.6 monolayer/cycle.

Grown films are $W_{1-x}N_x$ with $x=15-22\text{at}\%$ Resistivity of the deposited films was about 340 $\mu\Omega\text{cm}$ with STDV of resistivity 2- 5 % for 10 nm films.

200 nm Cu metallization does not show any changes in resistance after the 500 °C test. Annealed 100 nm Cu layer has the same specific resistivity as 200nm 2.1

$\mu\Omega\text{cm}$. No interaction of Barrier with Cu occurred during the annealing test up to 500 °C.

REFERENCES

- [1] S. Bystrova, J. Holleman, P. H. Woerlee, R. A. M. Wolters, SAFE 2000.
- [2] Wei- Min Li et al., IEEE 2002, Electronic Components and Technology Conference.
- [3] K.- E. Elers et al., *Thin Solid Films*, 434 p. 94- 99, 2003.
- [4] J.W. Klaus et al., *Appl. Surf. Sci.* 162- 163, p. 479- 491, 2000.
- [5] J. L. Alay, et al., *Surface and Interface Analysis*, Vol. 17, p.373- 382, 1991